

12

STOICHIOMETRY

SECTION 12.1 THE ARITHMETIC OF EQUATIONS (pages 353–358)

This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

► Using Everyday Equations (pages 353–355)

1. How can you determine the quantities of reactants and products in a chemical reaction?

2. Quantity usually means the _____ of a substance expressed in grams or moles.
3. A bookcase is to be built from 3 shelves (Sh), 2 side boards (Sb), 1 top (T), 1 base (B), and 4 legs (L). Write a “balanced equation” for the construction of this bookcase.

► Using Balanced Chemical Equations (page 354)

4. Is the following sentence true or false? Stoichiometry is the calculation of quantities in chemical reactions. _____
5. Calculations using balanced equations are called _____.

► Interpreting Chemical Equations (pages 356–357)

6. From what elements is ammonia produced? How is it used?

7. Circle the letter of the term that tells what kind of information you CANNOT get from a chemical equation.

a. moles

d. volume

b. mass

e. number of particles

c. size of particles

Stoichiometry Section 9 The Arithmetic Of Equations Answers

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General College Chemistry Charles William Keenan, Donald C. Kleinfelter, Jesse Hermon Wood, 1980 **Chemistry** Therald Moeller, 2012-12-02 Chemistry with Inorganic Qualitative Analysis is a textbook that describes the application of the principles of equilibrium represented in qualitative analysis and the properties of ions arising from the reactions of the analysis This book reviews the chemistry of inorganic substances as the science of matter the units of measure used atoms atomic structure thermochemistry nuclear chemistry molecules and ions in action This text also describes the chemical bonds the representative elements the changes of state water and the hydrosphere which also covers water pollution and water purification Water purification occurs in nature through the usual water cycle and by the action of microorganisms The air flushes dissolved gases and volatile pollutants when water seeps through the soil it filters solids as they settle in the bottom of placid lakes Microorganisms break down large organic molecules containing mostly carbon hydrogen nitrogen oxygen sulfur or phosphorus into harmless molecules and ions This text notes that natural purification occurs if the level of contaminants is not so excessive This textbook is suitable for both chemistry teachers and students **Computers &**

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Main and Advanced Chemistry Dr. K. G. Ojha, 2022-07-02 The new edition of IIT JEE Main Advanced CHEMISTRY is designed to present a whole package of Chemistry study preparation sufficing the requirements of the aspirants who are preparing for the upcoming exam Highlights of the Book Exam Pattern and Chemistry Syllabus for JEE Main and Advanced included An Analysis of IIT JEE included Chapter wise Theory detailed with 1000 examples 5000 Chapter wise Multiple Choice Questions 2500 Chapter wise Different Format Questions Chapter wise Assessment Test Chapter wise HOTS Problems Appendix on Equations Glossary JEE Main and Advanced Mock Test NEET Mock Test Answers to Questions included with Explanations Presence of accurate Diagrams and Tables From food to pharmaceuticals Chemistry plays a huge role in making informed decisions Therefore this book proves a comprehensive resource of Chemistry and serves to be a suitable Study Guide for the aspirants with focus on Qualitative Preparation and Systematic understanding of the Syllabus and Examination Level With provision for self assessment in Mock Tests this book stands beneficial in imprinting concepts in the mind **Chemistry**

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