12

# STOICHIOMETRY

#### SECTION 12.1 THE ARITHMETIC OF EQUATIONS (pages 353-358)

This section explains how to calculate the amount of reactants required or product formed in a nonchemical process. It teaches you how to interpret chemical equations in terms of interacting moles, representative particles, masses, and gas volume at STP.

#### ▶ Using Everyday Equations (pages 353-355)

- How can you determine the quantities of reactants and products in a chemical reaction?
- 2. Quantity usually means the \_\_\_\_\_\_ of a substance expressed in grams or moles.
- A bookcase is to be built from 3 shelves (Sh), 2 side boards (Sb), 1 top (T), 1 base (B), and 4 legs (L). Write a "balanced equation" for the construction of this bookcase.

#### ► Using Balanced Chemical Equations (page 354)

- Is the following sentence true or false? Stoichiometry is the calculation of quantities in chemical reactions.
- 5. Calculations using balanced equations are called

# ► Interpreting Chemical Equations (pages 356–357)

- 6. From what elements is ammonia produced? How is it used?
- Circle the letter of the term that tells what kind of information you CANNOT get from a chemical equation.
  - a. moles

d. volume

b. mass

e. number of particles

c. size of particles

# **Stoichiometry Section 12 1 The Arithmetic Of Equations**

**Jessica J Manson** 

**Stoichiometry Section 12 1 The Arithmetic Of Equations:** 

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