Name: __Chelsey Williams_____ Diabe: 3/31/22

Student Exploration: Stoichiometry

Vocabulary: Avogadro's number, balanced equation, cancel, coefficient, conversion factor, dimensional analysis, formula mass, molar mass, mole, molecular mass, stoichiometry

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- A 250 mL glass of orange juice contains 22 grams of sugar. How much sugar is in a two-liter (2,000 mL) bottle of orange juice? 176 grams
- It requires two sticks of butter to make a batch of 20 cookies. How much butter will it take to make 150 cookies? __15 sticks of butter

Gizmo Warm-up

Just as a cook follows a recipe to decide how much of each ingredient to add, a chemist uses stoichiometry to determine the amounts of substances involved in chemical reactions. The Stoichiometry Gizmo allows you to try your hand at figuring out the amounts of reactants and products that take part in a chemical reaction.

To begin, check that this equation is shown:

 Look at the coefficients (such as the "3" in 3CO) in front of each substance in the equation. The coefficients tell you how many molecules or atoms take part in a chemical reaction. In the spaces below, list the number of each molecule or atom in the equation:

New equation

New question

(III) oxide (Fe, Oxid)

Problems solved: 0

Fe₂O₃ (s) + 3CO (p) -- 2Fe (s) + 3CO₂ (p)

How many moles of carbon monoxide (CC) are

required to react completely with 1.75 motes of iron

In a balanced equation, the same number of each kind of atom is shown on each side of the equation. Calculate the number of iron (Fe), oxygen (O), and carbon atoms (C).

Stoichiometry Pg 113 Answers

Richard M. Felder, Ronald W. Rousseau, Lisa G. Bullard

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Pp/Chemistry BarCharts, Inc., 2008-06-18 , GATE Question Bank - Biotechnology Mocktime Publication, 2400 MCQs GATE Biotechnology Chapterwise Question Bank Based on New Syllabus

Elementary Principles of Chemical Processes Richard M. Felder, Ronald W. Rousseau, Lisa G. Bullard, 2020-08-11 This best selling text prepares students to formulate and solve material and energy balances in chemical process systems and lays the foundation for subsequent courses in chemical engineering The text provides a realistic informative and positive introduction to the practice of chemical engineering Practice Makes Perfect Chemistry Marian DeWane, Heather Hattori, 2011-06-03 Don't be confused by chemistry Master this science with practice practice Practice Makes Perfect chemistry is a comprehensive guide and workbook that covers all the basics of chemistry that you need to understand this subject Each chapter focuses on one major topic with thorough explanations and many illustrative examples so you can learn at your own pace and really absorb the information You get to apply your knowledge and practice what you ve learned through a variety of exercises with an answer key for instant feedback Offering a winning formula for getting a handle on science right away Practice Makes Perfect chemistry is your ultimate resource for building a solid understanding of chemistry fundamentals Prentice Hall Chemistry ,2000 How to Prepare for the SAT II Joseph A. Mascetta,1998 An extensive review of chemistry topics a full length diagnostic test precedes four up to date full length model SAT II Chemistry Tests All tests are followed with answers charts students can use to diagnose their individual study needs Chemistry topics reviewed include atomic structure bonding all other topics included in the SAT II test Joseph A Mascetta is a science educational consultant a former principal chemistry teacher at Mount Lebanon High School Pittsburgh PA **Programmed Solving Problems in Chemistry** Rod O'Connor, Charles **Topics in General Chemistry** Armine D. Paul, 1971 Mickey, Alton Hassell, 1977 Microbial Responses to Environmental Changes Jürg B. Logue, Stuart E. G. Findlay, Jérôme Comte, 2016-01-20 Advances in next generation sequencing technologies omics and bioinformatics are revealing a tremendous and unsuspected diversity of microbes both at a compositional and functional level Moreover the expansion of ecological concepts into microbial ecology has greatly advanced our comprehension of the role microbes play in the functioning of ecosystems across a wide range of biomes Super imposed on this new information about microbes their functions and how they are organized environmental gradients are changing rapidly largely driven by direct and indirect

human activities In the context of global change understanding the mechanisms that shape microbial communities is pivotal to predict microbial responses to novel selective forces and their implications at the local as well as global scale One of the main features of microbial communities is their ability to react to changes in the environment Thus many studies have reported changes in the performance and composition of communities along environmental gradients However the mechanisms underlying these responses remain unclear It is assumed that the response of microbes to changes in the environment is mediated by a complex combination of shifts in the physiological properties single cell activities or composition of communities it may occur by means of physiological adjustments of the taxa present in a community or selecting towards more tolerant better adapted phylotypes Knowing whether certain factors trigger one many or all mechanisms would greatly increase confidence in predictions of future microbial composition and processes This Research Topic brings together studies that applied the latest molecular techniques for studying microbial composition and functioning and integrated ecological biogeochemical and or modeling approaches to provide a comprehensive and mechanistic perspective of the responses of micro organisms to environmental changes This Research Topic presents new findings on environmental parameters influencing microbial communities the type and magnitude of response and differences in the response among microbial groups and which collectively deepen our current understanding and knowledge of the underlying mechanisms of microbial structural and functional responses to environmental changes and gradients in both aquatic and terrestrial ecosystems The body of work has furthermore identified many challenges and questions that yet remain to be addressed and new perspectives to follow up on Chemical principles Paul B. Kelter, 1998 **Elements of Experimental Organic Chemistry** Michael McCormick,2002 Cumulated Index Medicus ,1974 **Proteins** Ruth Hogue Angeletti, 1998-09-14 Proteins Analysis and Design focuses solely on individual experimental approaches rather than on specific classes of proteins The book provides insight into the important issues in protein science and how one can cope with them These include all issues which explore the detailed relationship of protein structure to function Provides problems and technical solutions Includes posttranslational modifications Uses synthetic peptides as biological models Details mutagenesis and protein engineering Covers design of protein structure and function Cehmistry Textbook for College and University USA Ibrahim Sikder, 2023-06-04 Cehmistry Textbook USA Chemical Abstracts ,1990 Student's Guide to Introduction to Chemical Principles by Edward I. Peters, 2d Ed Peter P. Berlow, 1978 Monthly List of Russian Accessions ,1957 Sci-tech Book Profiles ,1965 Includes title page table of contents list of contributors preface and all indexes of each book

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