

Stoichiometry

Mole-Mole Stoichiometry

Directions. Solve the following problems. Write an organized solution and box your final answer.

1. How many moles of hydrogen, H_2 , are needed to react with 2.0 moles of nitrogen, N_2 ?



2. How many moles of oxygen are produced by the decomposition of 6.0 moles of potassium chlorate, $KClO_3$?



3. How many moles of hydrogen are produced from the reaction of 3.0 moles of zinc?



4. How many moles of oxygen are necessary to react completely with 4.0 moles of propane (C_3H_8)?



5. How many moles of potassium nitrate are produced when 2.5 moles of potassium phosphate react?



Mass-Mole and Mass-Mass Stoichiometry

Directions. Solve the following problems. Write an organized solution and box your final answer.

1. How many moles of HCl will be produced when 249 g of $AlCl_3$ are reacted according to this chemical equation?



2. How many grams of NH_3 will be produced when 33.9 mol of H_2 are reacted according to this chemical equation?



3. Determine the number of grams of SO_3 that can be produced by the reaction of 45.3 g of SO_2 and O_2 :



4. What mass of Mg will be produced when 86.4 g of K are reacted?



5. What mass of H_2 will be produced when 122 g of Zn are reacted?



Stoichiometry Mole Mole Problems Key

Christian G. Meyer

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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