

Stoichiometry of Airbags

When automobile airbags are activated in a collision, they rapidly inflate with gas. This is achieved with a chemical reaction, a decomposition reaction, where a small amount of solid sodium azide (NaN_3) rapidly decomposes to sodium metal and nitrogen gas:



It is the nitrogen gas which fills the airbag. A **very** precise amount of sodium azide needs to be placed inside the airbag during the manufacturing process. Using your knowledge of stoichiometry, explain why you think it is important to use such a precise quantity of sodium azide.

Stoichiometry In Airbags

**John C. Kotz, Paul Treichel, Patrick A.
Harman**



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Stoichiometry In Airbags Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Stoichiometry In Airbags**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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