

Pre Lab:

Before you begin the lab, you must calculate the equivalent weight of the 5 ammonium salts listed in table 1. See below for some hints on how to complete the table:

Table 1: Equivalent Weight of the Five Possible Ammonium Salts

Salt	Formula	Molar Mass (g/mol)	Number of equivalents (eq/mol)	Equivalent weight (g/eq)
Ammonium Bromide	NH_4Br	97.94 g/mol	1 eq/mol	97.94 g/eq
Ammonium Chloride	NH_4Cl	53.49 g/mol	1 eq/mol	53.49 g/eq
Ammonium Iodide	NH_4I	144.94 g/mol	1 eq/mol	144.94 g/eq
Ammonium Nitrate	NH_4NO_3	80.05 g/mol	1 eq/mol	80.05 g/eq
Ammonium Sulfate	$(\text{NH}_4)_2\text{SO}_4$	132.15 g/mol	2 eq/mol	66.08 g/eq

Unknown Salt Identification

Jianjun Gao

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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College Practical Chemistry V K Ahluwalia, Sunita Dhingra, Sunita Dhingra, 2005 , **A Method for the Identification of Pure Organic Compounds by a Systematic Analytical Procedure Based on Physical Properties and Chemical Reactions ...** Samuel Parsons Mulliken, 1922 **Understanding 'O' Level Chemistry through Problem Solving** , *Breakthrough to CLIL for Chemistry Workbook* Richard Harwood, Timothy Chadwick, 2015-04-30 A series of workbooks offering integrated content and language support for specific subjects Breakthrough to CLIL for Chemistry Age 14 helps ESL EAL students get the most out of their studies when learning subjects through the medium of English The workbook contains exercises set within the context of core topics to consolidate understanding embedding practice in aspects of language central to the subject in question It is designed to support any Chemistry curriculum for students aged 14 16 including UK GCSE Cambridge IGCSE and IB MYP The book should be used alongside a core textbook and may be used within the classroom or as a self study or homework resource **Salts and Their Reactions** Leonard Dobbin, Hugh Marshall, 1904 *Advanced Inorganic Analysis - I* Mr. Ajeet Singh, 2018-05-22 This book provides in depth coverage of modern methods in inorganic analysis including advanced qualitative and quantitative techniques instrumentation and interpretation of results catering to postgraduate and research level chemistry students **Lab Manual for Investigating Chemistry** David Collins (Ph. D.), Matthew Johll, 2008-12-02 While many of the core labs from the first edition have been retained a renewed focus on the basics of chemistry and the scientific process create an even more detailed supplemental offering **ISC Practical Chemistry Vol. I Class-XI** S.P. Sharma & Dr. Ajaya Baboo, Across All Boards ICSE ISC Boards Abiotic stress: molecular genetics and genomics, Volume II Mukesh Jain, Rohini Garg, Rajeev K. Varshney, Prasanta Kumar Subudhi, *EduGorilla's CBSE Class 11th Chemistry Lab Manual | 2024 Edition | A Well Illustrated, Complete Lab Activity book with Separate FAQs for Viva Voce Examination* , Need an informative and well illustrated Lab Manual CBSE Class 11th Chemistry Lab Manual is here for you The Lab Manual provides comprehensive steps for guiding students through each experiment Rigorously researched content prepared by a team of educators writers editors and proofreaders CBSE Class XI Chemistry Lab Manual has properly labeled high resolution diagrams and graphs A separate section on Viva Questions has been included to aid students in their Viva examination The Lab Manual explains the complex topics through detailed illustrations and lucid language making them simple to grasp Worksheets have been provided in CBSE Class 11th Chemistry Lab Manual for doing rough work Modern Experimental Chemistry George W. Jr. Latimer, 2012-12-02 Modern Experimental Chemistry provides techniques of qualitative analysis that reinforce experiments on ionic equilibria This book includes the determination of water in hydrated salts identification of an organic compound after determining its molecular weight and nonaqueous titration of a salt of a weak acid The calculation of chemical stoichiometry calculation of thermodynamic properties by determining the change in equilibrium with temperature and chromium chemistry are also

covered This compilation contains enough experiments for classes which have six hours of laboratory two 3 hour meetings per week to last two semesters This publication is intended for chemistry students as an introductory manual to chemistry laboratory *Laboratory Manual for Chemistry Fundamentals* Phyllis Buell,James Girard,2002-08 Physical Sciences

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