

Name _____

Surface Tension Science Lab

Problem: Do different chemicals affect surface tension? Will surface tension be different when water is mixed with salt or detergent?

Possible Topics to Be Researched for the Review of Literature:

- surface tension
- chemical structure of water
- covalent bond
- hydrogen bonds

Hypothesis: I think _____

Materials: Pennies, cups, water, salt, detergent, measuring spoons, paper towels, eye droppers, data sheet, graduated cylinders

Procedure:

1. Fill one cup with exactly 50 mL of water. Label the cup "Water."
 - Put one penny on a paper towel.
 - Using an eye dropper, add one drop at a time from the "Water" cup until the water spills off of the penny. Count the drops accurately as you put them on the penny.
 - Wipe the penny off, then repeat the test 9 more times for a total of 10 trials.
 - Record the count for each trial (See Data Sheet)
2. Fill a second cup with exactly 50 mL of water. Add 1 teaspoon of salt and stir. Label the cup "Water with Salt."
 - Put one penny on a paper towel.
 - Using an eye dropper, add one drop at a time from the "Water with Salt" cup until the water spills off of the penny. Count the drops accurately as you put them on the penny.
 - Wipe the penny off, then repeat the test 9 more times for a total of 10 trials.
 - Record the count for each trial (See Data Sheet)
3. Fill a third cup with exactly 50 mL of water. Add 1 teaspoon of liquid detergent and stir. Label the cup "Water with Detergent."
 - Put one penny on a paper towel.
 - Using an eye dropper, add one drop at a time from the "Water with Detergent" cup until the water spills off of the penny. Count the drops accurately as you put them on the penny.
 - Wipe the penny off, then repeat the test 9 more times for a total of 10 trials.
 - Record the count for each trial (See Data Sheet)

Surface Tension Lab Report

ML Morrison

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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