

SUPERSATURATED SOLUTIONS: ROCK CANDY LAB

Purpose: To introduce crystal growth in order to demonstrate the properties of supersaturated solutions

Information: Solubility of substances improves with stirring (mechanical energy) and by heating (heat energy). You will find that the ***solvent*** dissolves only so much ***solute***. After a while, the excess solute precipitates to the bottom. When this happens, the liquid is a ***saturated solution***.

Solutes dissolve in larger quantities when placed into heated liquids (solvents) because the molecules of the heated liquids vibrate further apart. The spaces between the molecules

become greater and more solid molecules (sugar) can fill these spaces. As the liquid cools

down, the spaces between the molecules become smaller and the excess solids precipitate to

the bottom. Heating solvents to dissolve additional solute. Once the solution cools, it now has

more solute than it normally could have, and this is called a ***supersaturated solution***.

Once a solution becomes supersaturated, it wants to crystallize the excess solute. The solute will form crystals on any surface. In our case the surface is a string, and we are using a

lifesaver candy as our "seed." A seed is a starting point, a solid piece of sugar candy will mimic a sugar crystal, and start the chain reaction of crystallization. If the jar is left uncovered, crystals may even form on the dust that settles on top.

Equipment: 250 mL beaker, scale, sugar, water, a clean jar, string, stirring rod, and candy

Procedure:

1. Using the 250 mL beaker, measure 150 mL of water.

2. Measure out 450 g of Sugar using a plastic cup and spoon.

3. Adding 1 spoonful of sugar at a time, dissolve as much sugar as possible in the water. (you may add food coloring at this time if desired)

4. Once you have a saturated solution, place the beaker on a hotplate and allow the water to

get hot, but do not bring to a boil

5. While waiting for the water, prepare a string by measuring approximately 15 cm and tying

one to a Lifesaver.

6. Once the solution is hot, allow your instructor to bring it to your lab desk. Add one spoonful

at a time of sugar until no more sugar can be dissolved. If the solution cools too much, reheat it on the hotplate.

7. Stir to help dissolve the sugar. When no more sugar dissolves and some remains in the

Supersaturated Solutions Rock Candy Lab Answers

Kathy Dodds Tyner



Supersaturated Solutions Rock Candy Lab Answers:

DeGarmo's Materials and Processes in Manufacturing Ronald A. Kohser, Peter C. Collins, J. T. Black, 2024-12-31

Classic textbook introducing key concepts in manufacturing with a focus on practical applications updated to include the latest industry developments For over 65 years DeGarmo's Materials and Processes in Manufacturing has comprehensively presented both traditional and new manufacturing materials processes and systems in a descriptive non mathematical manner Students are first introduced to a range of engineering materials including metals plastics and polymers ceramics and composites The processes used to convert this stuff into things are then described along with their typical applications capabilities and limitations Segments cover casting forming machining welding and joining and additive manufacturing Supporting chapters present concepts relating to material selection heat treatment surface finishing measurement inspection and manufacturing systems The Fourteenth Edition has been updated to reflect the most current technologies Coverage of additive manufacturing 3D printing has been significantly expanded along with updates on new and advanced materials Case studies are featured throughout the book and review problems have been placed at the end of each chapter A full collection of online bonus material is provided for both students and instructors DeGarmo's Materials and Processes in Manufacturing Fourteenth Edition includes information on Equilibrium phase diagrams and the iron carbon system heat treatment and process capability and quality control Expendable mold and multiple use mold casting processes powder metallurgy particulate processing fundamentals of metal forming and bulk forming and sheet forming processes Cutting tool materials turning and boring processes milling drilling and related hole making processes and CNC processes and adaptive control in the A 4 and A 5 levels of automation Sawing broaching shaping and filing machining processes thread and gear manufacturing and surface integrity and finishing processes DeGarmo's Materials and Processes in Manufacturing has long set the standard for introducing students to the materials and processes in product manufacturing and has been incorporated in programs of manufacturing mechanical industrial metallurgical and materials engineering as well as various technology degrees Its descriptive nature provides an excellent first exposure to its various subjects which may then be followed by advanced courses in specific areas

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to material behavior and selection measurement and inspection machining fabrication molding fastening and other important processes using plastics ceramics composites and ferrous and nonferrous metals and alloys This extensive overview of the field gives students a solid foundation for advanced study in any area of engineering manufacturing and technology

Science I Essential Interactions ,2000-10

Laboratory Exercises for Preparatory Chemistry Kathy Dodds

Tyner,1994-06 Laboratory Exercises for Preparatory Chemistry is the perfect complement to a one semester preparatory chemistry laboratory course Tyner s manual emphasizes the application of chemistry and the principles of science to everyday life The labs are directly applicable to the real world and often contain supplemental assignments that illustrate an application Teaching Mineralogy John Brady,David W. Mogk,Dexter Perkins,1997

Popular Science Monthly and World

Advance ,1932

Exploring Chemistry in Today's World Kathy L. Tyner,1993 The labs were specifically chosen with

several goals in mind a To parallel lecture topics b To demonstrate important chemical principles c To employ the use of techniques of self discovery and the scientific method d To illustrate topics that are of public interest or concern e To encourage the application of chemistry outside the laboratory In keeping with these goals the author has included laboratory assignments that are applicable to the real world or contain supplemental exercises that illustrate an application Where possible commercial products are used such as aspirin antacids etc Each lab begins with written objectives Then in an effort to increase involvement before the lab work begins questions are posed that ask the student a To make predictions about the outcome of the experiment b To formulate a hypothesis c To think about a phenomenon in a specific way d To apply personal experience in answering a questions Pref Popular Science Monthly ,1933

Exploring Chemistry (Loose-Leaf)

Matthew Johll,2012-05-04 Matthew Johll s Exploring Chemistry overs the standard topics for the nonmajors course in the typical order but each chapter unfolds in the context of a single case study that helps students connect what they are learning to real life situations For example students work through the often difficult topics of molecular structure gas laws and organic chemistry by learning about the development of powerful new chemotherapy drugs new technologies for screening airline passengers and the creation of biodegradable biopolymers It s the same same case driven approach that Johll uses in his acclaimed Investigating Chemistry now in its Third Edition but Exploring Chemistry goes beyond the other book s specific focus on examples from forensic science to use real life stories from cooking athletics genetics green chemistry and more

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