### Intext Questions

legion 1. Classify the following amines as primary, secondary or

Primary amine contains —NH<sub>2</sub> group; secondary amine NH group; bettiary amine N group.

ibifion. (i) Primary (ii) Tertiary

(iii) Primary (iv) Secondary

bestion 2. (i) Write the structures of different isomeric amines assessed to the molecular formula, C<sub>4</sub>H<sub>11</sub>N.

- (ii) Write IUPAC names of all the isomers.
- (iii) What type of isomerism is exhibited by different pairs of amines?
  - (i) Recall the definition of isomerism; rules of IUPAC nomenclature; types of isomerism.
  - (ii) Draw all the structures possible for molecular formula C<sub>4</sub>H<sub>1</sub>N and give their IUPAC name according to rules.
  - (iii) Also compare all the structures to find the property like chain length, position of functional group, nature of functional group etc. in which the structures differ.

fultion. (i) and (ii) Eight isomers of C4H11N are:

(b) 
$$\stackrel{\circ}{C}H_3 \stackrel{\circ}{-}\stackrel{\circ}{C}H \stackrel{\circ}{-}\stackrel{\circ}{C}H_2 - NH_2$$
2-methylpropan-1-amine
(Primary)
 $\stackrel{\circ}{C}H_3$ 

(c) 
$$CH_3$$
 $CH_3$ 
 $CH_3$ 

d) 
$$CH_3 - CH_3$$
 $CH_3 - CH_3$ 

2-methylpropan-2-amine (Primary)

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