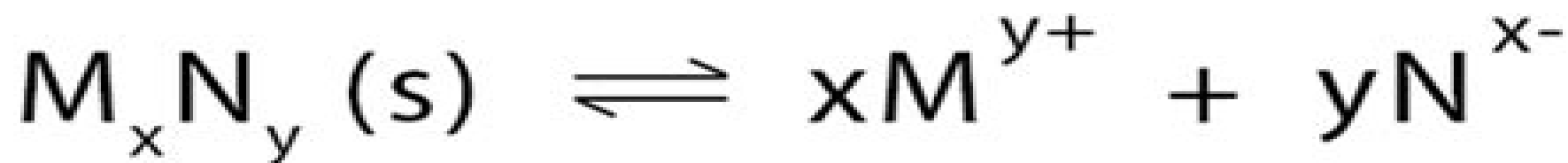


# Solubility Product Constant ( $K_{sp}$ )



$$K_{sp} = [M^{y+}]^x [N^{x-}]^y$$

$[M^{y+}] \rightarrow$  Concentration of  $M^{y+}$

$[N^{x-}] \rightarrow$  Concentration of  $N^{x-}$

# Solubility Product Pogil

**Gilbert F. Smith**



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