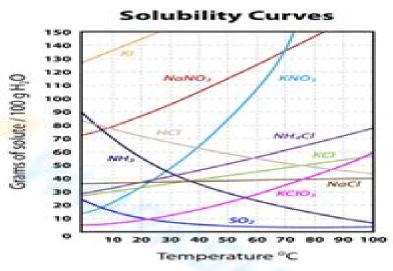
WS 6.1: Interpreting Solubility Curves

Directions: Use the Solubility Curve Table below to answer all questions. So you do not have to keep flipping your paper over you might want to put the solubility curve on your phone or computer screen when answering the questions on the back. All questions assume the amount of water is 100 ml.



What is the solvent in this Solubility Curve? water

80°C = -146 g

- Saturation points are given per how much solvent? 100 ml or 100 g.
- 3) What is the saturation point of NaNOs at 10°C, 40°C, & 80°C?
 # 10°C = -79 g
 # 40°C = -105 g
- How much NHs can you dissolve in water at 10°C, 30°C, & 90°C?
 10°C = -70 g
 30°C = -45 g
 90°C = -10 g
- 5) What is the maximum amount of KCl that can be dissolved in water at 20°C, 50°C, & 90°C? # 20°C = -33 g # 50°C = -42 g # 90°C = -53 g

6) If you want to make a saturated solution of NaCl how many grams would you dissolve in 100 grams of water # 80°C?

About 40 grams

7) Except for KI What substance is the most soluble in water at 20°C?

NaNO

 Except for KL what substance is the most soluble in water at 72°C?

KINDO.

- 9) Look up The boiling points for SO₂, HCl, and NH₃ and write them below:
 B.P. SO₂ = -10°C
 B.P. HCl = -85°C
 B.P. NH₃ = -33°C
- 10) Based on the boiling points you looked up, are SO₂, HCl, & NH₃ solids, liquids, or gases at 0°C? (and anywhere else on the solubility curve) gases

Solubility Curves Practice Answer

The Princeton Review

Solubility Curves Practice Answer:

Unveiling the Power of Verbal Artistry: An Emotional Sojourn through Solubility Curves Practice Answer

In a world inundated with displays and the cacophony of instantaneous connection, the profound energy and mental resonance of verbal artistry often diminish in to obscurity, eclipsed by the continuous onslaught of noise and distractions. Yet, set within the musical pages of **Solubility Curves Practice Answer**, a fascinating perform of literary elegance that impulses with fresh thoughts, lies an unforgettable trip waiting to be embarked upon. Composed by way of a virtuoso wordsmith, that exciting opus courses visitors on a psychological odyssey, lightly revealing the latent possible and profound affect embedded within the elaborate web of language. Within the heart-wrenching expanse of the evocative examination, we can embark upon an introspective exploration of the book is main themes, dissect its charming publishing type, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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