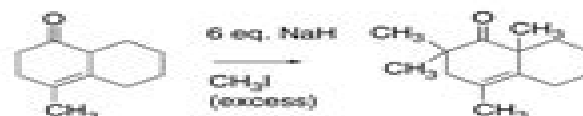


- b. This transformation involves methylation at all enolizable positions. The alkylation was effected using a sixfold excess of NaH and excess methyl iodide. Evidently there is not a significant amount of methylation at C(4), which could occur through  $\gamma$ -alkylation of the C(8a)-enolate.



- c. This alkylation was accomplished using two equivalents of  $\text{NaNH}_2$  in liquid  $\text{NH}_3$ . The more basic site in the dianion is selectively alkylated. Note that the dianion is an indenyl anion, and this may contribute to its accessibility by di-deprotonation.



- d. This is a nitrile alkylation involving an anion that is somewhat stabilized by conjugation with the indole ring. The anion was formed using  $\text{NaNH}_2$  in liquid  $\text{NH}_3$ .



- e. The structure of the anion is shown in the reaction scheme. The anion is formed by deprotonation of the nitrile group of 1-methyl-1H-indole-3-carbonitrile with  $\text{NaNH}_2$  in liquid  $\text{NH}_3$ . The anion is then alkylated with  $\text{PhCH}_2\text{Cl}$  to form the final product, 1-methyl-2-benzyl-1H-indole-3-carbonitrile.
- f. The structure of the anion is shown in the reaction scheme. The anion is formed by deprotonation of the nitrile group of 1-methyl-1H-indole-3-carbonitrile with  $\text{NaNH}_2$  in liquid  $\text{NH}_3$ . The anion is then alkylated with  $\text{PhCH}_2\text{Cl}$  to form the final product, 1-methyl-2-benzyl-1H-indole-3-carbonitrile.



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**Lauren Gardner**



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