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Solubility Curve Practice Problems Worksheet 1

You'll notice that for most substances, solubility increases as temperature increases. As discussed earlier in solutions involving liquids and solids typically more solute can be dissolved at higher temperatures. Can you find any

exceptions on the graph?_____

Here's an example of how to read the graph. Find the curve for KClOs.

At 30°C approximately 10g of KClO₃ will dissolve in 100g of water. If the temperature is increased to 80°C, approximately ______ of the substance will dissolve in 100g (or 100mL) of water.

Directions: Use the graph to answer the following questions. REMEMBER UNITS!

- What mass of solute will dissolve in 100mL of water at the following temperatures?
 - a. KNO3 at 70°C = ______
 - b. NaCl at 100°C= _____
 - c. NH₄Cl at 90°C=
 - d. Which of the above three substances is most soluble in water at 15°C. =



Solubility Curve Practice Problems

JA Banks

Solubility Curve Practice Problems:

Reviewing **Solubility Curve Practice Problems**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Solubility Curve Practice Problems**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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