

SOLUBILITY OF COMMON COMPOUNDS IN WATER

The term soluble here means > 0.1 mol/L at 25°C.

Negative Ions (Anions)	Positive Ions (Cations)	Solubility of Compounds
All	Alkali ions: Li^+ , Na^+ , K^+ , Rb^+ , Cs^+ , Fr^+	Soluble
All	Hydrogen ion: H^+	Soluble
All	Ammonium ion: NH_4^+	Soluble
Nitrate, NO_3^-	All	Soluble
Chloride, Cl^- or Bromide, Br^- or Iodide, I^-	All others	Soluble
	Ag^+ , Pb^{2+} , Cu^+	Low Solubility
Sulphate, SO_4^{2-}	All others	Soluble
	Ag^+ , Ca^{2+} , Sr^{2+} , Ba^{2+} , Pb^{2+}	Low Solubility
Sulphide, S^{2-}	Alkali ions, H^+ , NH_4^+ , Be^{2+} , Mg^{2+} , Ca^{2+} , Sr^{2+} , Ba^{2+}	Soluble
	All others	Low Solubility
Hydroxide, OH^-	Alkali ions, H^+ , NH_4^+ , Sr^{2+}	Soluble
	All others	Low Solubility
Phosphate, PO_4^{3-} or Carbonate, CO_3^{2-} or Sulphite, SO_3^{2-}	Alkali ions, H^+ , NH_4^+	Soluble
	All others	Low Solubility

Solubility Chart Chemistry 12

KJ Lindholm-Leary



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New tables Section 8 Analytical Chemistry Figures of Merit Common Symbols Used in Gas and Liquid Chromatographic Schematic Diagrams Varieties of Hyphenated Gas Chromatography with Mass Spectrometry Section 15 Practical Laboratory Data Standard Fittings for Compressed Gas Cylinders Plug and Outlet Configurations for Common Laboratory Devices Section 16 Health and Safety Information Abbreviations Used in the Assessment and Presentation of Laboratory Hazards Incompatible Chemicals Explosion Shock Hazards Water Reactive Chemicals Testing Requirements for Peroxidizable Compounds Tests for the Presence of Peroxides Pyrophoric Compounds Compounds That Are Reactive with Air Flammability Hazards of Common Solvents Selection of Laboratory Gloves Selection of Respirator Cartridges and Filters Selection of Protective Laboratory Garments Protective Clothing Levels Chemical Fume Hoods and Biological Safety Cabinets Gas Cylinder Safety and Stamped Markings Laser Hazards in the Laboratory General Characteristics of Ionizing Radiation for the Purpose of Practical Application of Radiation Protection Radiation Safety Units Significantly updated and expanded tables Section 1 Basic Constants Units and Conversion Factors Update of Standard Atomic Weights 2013 Update of Atomic Masses and Abundances Section 8 Analytical Chemistry Expansion of Abbreviations and Symbols Used in Analytical Chemistry Section 9 Molecular Structure and Spectroscopy Update of Bond Dissociation Energies Section 12 Properties of Solids Major update and Expansion of Electron Stopping Powers Section 14 Geophysics Astronomy and Acoustics Major Update of Interstellar Molecules Update of Atmospheric Concentration of Carbon Dioxide 1958 2013 Update of Global Temperature Trend 1880 2013 Section 15 Practical Laboratory Data Major update of Reference Points on the ITS 90 Temperature Scale Update of Laboratory Solvents and Other Liquid Reagents Section 16 Health and Safety Information Update of Flammability

of Chemical Substances Update of Threshold Limits for Airborne Contaminants to 2013 values Appendix B Update of Sources of Physical and Chemical Data *CRC Handbook of Chemistry and Physics, 94th Edition* William M. Haynes, 2016-04-19

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