#### SOLUBILITY CURVE WORKSHEET KEY

Use your solubility curve graphs provided to answer the following questions.

- What are the customary units of solubility on solubility curves? <u>Degress Celsius and</u> grams of solute/100g of water
- Define solubility. A measure of how much solute can dissolve in a given amount of solvent.
- According to the graph, the solubility of any substance changes as <u>temperature</u> changes.
- List the substances whose solubility decreases as temperature increases. NH<sub>3</sub> and Ce2(SO4).
- Which substance is least affected by temperature changes? NaCl.
- How many grams of ammonium chloride (NH<sub>4</sub>CI) at 50°C? \_\_50q\_
- NaCl and KCIO<sub>3</sub> have the same solubility at approximately 78°C.
- Which compound is least soluble in water at 10°C? KCIO<sub>1</sub>
- How many grams of KNO<sub>3</sub> can be dissolved at 50°C? 80g
- 10. Are the following solutions unsaturated, saturated, or supersaturated?
  - a. 45g of NaNO<sub>3</sub> in 100 g of water at 30°C. saturated
  - b. 60g of KCIO<sub>3</sub> in 100 g of water at 90°C. <u>supersaturated</u>
- How many grams of sodium chloride, NaCl are required to saturate 100 grams of water at 100° C? 40g
- 12. How many grams of NaNO3 are required to saturate 100 grams of water at 75°C? 140g
- How many grams of KI will saturate water at 20°C? 33q
- 14. At what temperature would 25g of potassium chlorate (KCIO<sub>3</sub>) dissolve? 60°C
- At what temperature would 60g of NH<sub>4</sub>Cl dissolve? 70°C
- 16.89 g NaNO3 is prepared at 30°C.
  - a) Will all of the salt dissolve? No
  - b) What mass of NaNO<sub>3</sub> will dissolve at this temperature? 95q
- If 50 grams of NH<sub>4</sub>CI is dissolved at 50°C, how many additional grams NH<sub>4</sub>CI would be needed to make the solution saturated at 80°C? 15q
- 18. At 50°C, how many grams of KNO3 will dissolve? 80q
- At 70°C, how many grams of cerium (III) sulfate (Ce<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>) dissolve?
- Determine if each of the following is unsaturated, saturated, or supersaturated.
- a. 55g of NH<sub>3</sub> at 20°C supersaturated f. 78g of NaNO<sub>3</sub> at 10°C. saturated
- b. 10g of Ce<sub>2</sub>(SO<sub>4</sub>)<sub>1</sub> at 10°C unsaturated g. 145g of NaNO<sub>1</sub> at 80°C, saturated
- c. 110g of KNO<sub>3</sub> at 60°C. supersaturated h. 35g of NaCl at 100°C. unsaturated
- d. 65g of NH<sub>4</sub>Cl at 80°C. saturated
- e. 12g of NH<sub>3</sub> at 90°C. supersaturated

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