



Interpreting Solubility Curve



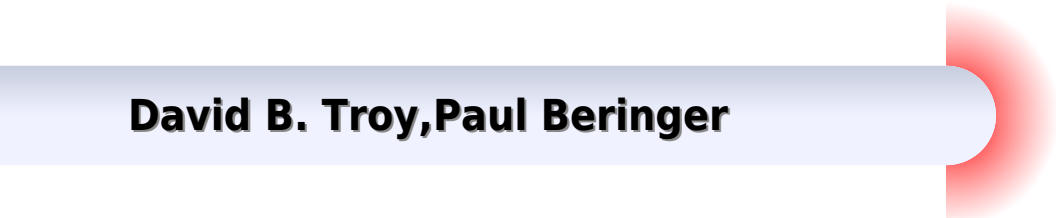
1. Define solubility. _____
2. Using the solubility curve below, answer the given questions.



- I. What is the solvent in this solubility curve? _____
- II. Saturation points are given per how much of the solvent? _____
- III. What is the saturation point of NaNO_3 at 10°C , 40°C , & 80°C ?
 10°C = _____ 40°C = _____ 80°C = _____
- IV. How much NH_3 can you dissolve in water at 10°C , 30°C , & 90°C ?
 10°C = _____ 30°C = _____ 90°C = _____
- V. Which salt is the least soluble in water? _____
- VI. How many grams of potassium chloride can be dissolved in 200 g of water at 80°C ? _____
- VII. At 40°C , how much potassium nitrate will be dissolved in 300 g of water? _____
- VIII. Which salt shows the least change in solubility from 0° to 100°C ? _____
- IX. Which substance shows a decrease in solubility from 0° to 100°C ? _____
- X. At 30°C , 90 g of sodium nitrate is dissolved in 100 g water. Is this solution saturated, unsaturated, or supersaturated? _____

Solubility Curve Practice Problems Packet Answers

David B. Troy, Paul Beringer



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 - Chapter 1: Introduction to Solubility Curve Practice Problems Packet Answers
 - Chapter 2: Essential Elements of Solubility Curve Practice Problems Packet Answers
 - Chapter 3: Solubility Curve Practice Problems Packet Answers in Everyday Life
 - Chapter 4: Solubility Curve Practice Problems Packet Answers in Specific Contexts
 - Chapter 5: Conclusion
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