



- [Index](#)
- [Language](#) •
- [General](#) •
- [Business](#) •
- [Science](#) •
- [Mathematics And Applied Science](#) •
- [Home Economics](#) •
- [Civil And Mechanical](#) •

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- [Subject Home](#)
- [1](#)
- [2](#)
- [3](#)
- [4](#)
- [5](#)

Menu

[General Comments](#)

[Weakness/Remedies](#)

[Candidate's Strength](#)

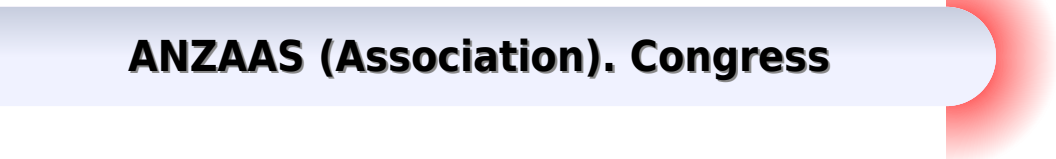
Question 1

- (a) (i) Define an acid according to the Lewis concept.
 (ii) Give one example of a Lewis acid. [3 marks]
- (b) Explain salting out in soap preparation. [2 marks]
- (c) State the reagent and condition necessary for the following conversion:

$$\text{H}-\text{C}\equiv\text{C}-\text{H} \rightarrow \text{Ag}-\text{C}\equiv\text{C}-\text{Ag}$$
 [2 marks]
- (d) What is the percentage abundance of an isotope? [2 marks]
- (e) (i) Why does the element with atomic number 18 not have an oxide?
 (ii) Explain why chlorine(I) oxide has a low melting point. [3 marks]
- (f) Describe a test to distinguish between concentrated HNO_3 and concentrated H_2SO_4 . [3 marks]
- (g) State two differences between an *electrochemical cell* and an *electrolytic cell*. [2 marks]
- (h) How does the trend in ionization energy affect the reactivity of group I element? [3 marks]
 Define the term *molecular formula*

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papers, 1881 Impact of Examination Policy on Teaching Chemistry in Nigerian Secondary Schools David Afolayan

Alao, 1988 **Daily Graphic** Ransford Tetteh, 2011-02-23 *West African Journal of Education*, 1957 MIJOLL, 2007-07

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Remember Practice is the key to success So practice more and more questions on each and every topic from a good question bank And here s the last but not the least and the most important tip revise all your incorrectly attempted questions repeatedly and bookmark important questions so that you gain perfection in attempting questions having similar concept or trick IT S NOT ALWAYS ABOUT STUDYING HARD BUT STUDYING SMART Regards Dr Monica Bhandari

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Donnelly,2020-01-28 Freesciencelessons is the most viewed GCSE Science channel on Youtube and has helped countless students to achieve or exceed their expectations in their GCSE Science exams This is the companion workbook for the GCSE Combined Science Foundation Chemistry 2 videos for the AQA specification Written by Shaun Donnelly who appears in every video this book contains hundreds of questions to guide students through the course Accompanied by detailed on line answers this is the perfect resource for every student following the AQA specification Every topic is covered in depth including all the required practicals A specimen exam paper is also included allowing students to check their progress across the subject

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Table of Contents Today Waec Question On Chemistry Paper 2

1. Understanding the eBook Today Waec Question On Chemistry Paper 2
 - The Rise of Digital Reading Today Waec Question On Chemistry Paper 2
 - Advantages of eBooks Over Traditional Books
2. Identifying Today Waec Question On Chemistry Paper 2
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Today Waec Question On Chemistry Paper 2
 - User-Friendly Interface
4. Exploring eBook Recommendations from Today Waec Question On Chemistry Paper 2
 - Personalized Recommendations
 - Today Waec Question On Chemistry Paper 2 User Reviews and Ratings
 - Today Waec Question On Chemistry Paper 2 and Bestseller Lists

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 - Today Waec Question On Chemistry Paper 2 Enhanced eBook Features
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 - Adjustable Fonts and Text Sizes of Today Waec Question On Chemistry Paper 2
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8. Staying Engaged with Today Waec Question On Chemistry Paper 2
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Today Waec Question On Chemistry Paper 2
9. Balancing eBooks and Physical Books Today Waec Question On Chemistry Paper 2
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Today Waec Question On Chemistry Paper 2
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
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 - Setting Reading Goals Today Waec Question On Chemistry Paper 2
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Today Waec Question On Chemistry Paper 2
 - Fact-Checking eBook Content of Today Waec Question On Chemistry Paper 2
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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