

CDX-G1200U/G1201U/ G1202U/G1280UM

SERVICE MANUAL

Ver. 1.0 2016.08



Photo: CDX-G1200U

The service manual of the mechanism deck, used in this model, has been issued in a separate volume. Please refer to the service manual of the MG-101 series for the mechanism deck information.

- The tuner and CD sections have no adjustments.

*US Model
Canadian Model
Australian Model
Hong Kong Model
Argentina Model
CDX-G1200U
AEP Model
UK Model
Saudi Arabia Model
CDX-G1200UM/G1201UM/G1202U
E Model
Korean Model
CDX-G1200UM/G1201U
Indian Model
CDX-G1200UM/G1280UM*

Model Name Using Similar Mechanism	CDX-G1100UM/G1150UM/G3100UP
Mechanism Type	MG-101CF-188
Optical Pick-up Name	DAX-25A

SPECIFICATIONS

GENERAL

Use the functions in the USA, NOT APPLICABLE IN CANADA, INCLUDING IN THE PROVINCE OF QUEBEC.
FROM USE QUANTITY AND CHARGE INFO, NOT APPLICABLE IN CANADA, IN CANADA (A. MODEL) (B. MODEL).

AUDIO POWER SPECIFICATIONS

CDX-G1200U
Power Output (RMS) (100 Hz, 1 kHz, 100 V)
100 W (100 V, 100 Hz)
100 W (100 V, 100 Hz)

FM SECTION

FM, 87.5-108 MHz

Tuning range: 87.5 MHz ~ 108 MHz
Antenna (external) connection
Automatic antenna (internal) connection
Intermediate frequency: 10.7 MHz
IF filter: 10.7 MHz (10.7 MHz) and 10.7 MHz (10.7 MHz)
Modulation: 100% (100%)
Sensitivity: 100 dB (100 dB)
Signal-to-noise ratio: 70 dB
Separation: 100 dB (100 dB)
Frequency response: 20 Hz ~ 20,000 Hz

AM

Tuning range: 530 kHz ~ 1,700 kHz
Antenna (external) connection
Automatic antenna (internal) connection
Sensitivity: 100 dB

CD SECTION

CD

Tuning range: 150 Hz ~ 1,000 Hz
Power Output (RMS) (100 Hz, 1 kHz, 100 V)
100 W (100 V, 100 Hz)
100 W (100 V, 100 Hz)
Modulation: 100% (100%)
Sensitivity: 100 dB (100 dB)
Signal-to-noise ratio: 70 dB
Separation: 100 dB (100 dB)
Frequency response: 20 Hz ~ 20,000 Hz

FM SECTION

Tuning range: 87.5 MHz ~ 108 MHz
Antenna (external) connection
Automatic antenna (internal) connection
Intermediate frequency: 10.7 MHz
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Sensitivity: 100 dB (100 dB)
Signal-to-noise ratio: 70 dB
Separation: 100 dB (100 dB)
Frequency response: 20 Hz ~ 20,000 Hz

AM

Tuning range: 530 kHz ~ 1,700 kHz
Antenna (external) connection
Automatic antenna (internal) connection
Sensitivity: 100 dB

CD

Tuning range: 150 Hz ~ 1,000 Hz
Power Output (RMS) (100 Hz, 1 kHz, 100 V)
100 W (100 V, 100 Hz)
100 W (100 V, 100 Hz)
Modulation: 100% (100%)
Sensitivity: 100 dB (100 dB)
Signal-to-noise ratio: 70 dB
Separation: 100 dB (100 dB)
Frequency response: 20 Hz ~ 20,000 Hz

CD

Tuning range: 150 Hz ~ 1,000 Hz
Power Output (RMS) (100 Hz, 1 kHz, 100 V)
100 W (100 V, 100 Hz)
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Modulation: 100% (100%)
Sensitivity: 100 dB (100 dB)
Signal-to-noise ratio: 70 dB
Separation: 100 dB (100 dB)
Frequency response: 20 Hz ~ 20,000 Hz

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US, Canadian, E, Argentina, Korean, Indian and Australian models
FM/AM COMPACT DISC PLAYER
AEP, UK and Hong Kong models
FM/MW/LW COMPACT DISC PLAYER
Saudi Arabia model
FM/MW/SW COMPACT DISC PLAYER

Sony Cdx Gt120 Car Receivers Owners Manual

Sony (Great Britain) Ltd



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web a at atmospheric pressure helium can exist in all three phases as well as a supercritical fluid phase near absolute zero choice b solid and gaseous helium never exist in equilibrium with each other at any temperature or pressure b solid and gaseous helium never exist in equilibrium with each other at any temperature or pressure choice c

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web stant is the foundation of thermochemistry it allows us to describe heat flow in terms of the change in a state function

enthalpy change is path independent note with regard to choice b that enthalpy change equals internal energy change only if the volume of the system remains constant and thus no thermodynamic work is performed 5 b

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web conceptual answers in order for the reaction to occur spontaneously ΔG for the reaction must be less than zero in this case ΔS must be positive and the $T\Delta S$ term outweighs the positive value of ΔH

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web 1 answer a explanation the value of ΔH is negative therefore the reaction is exothermic or heat releasing 2 answer a 1 explanation the entire reaction was multiplied by 2 therefore ΔH must also be halved 3 answer c explanation option c is incorrect because 566 kJ/mol is the ΔH per mole of O_2 and not for 2 moles

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web thermochemical equations equations that represent both mass and enthalpy changes $H_2O(s) \rightarrow H_2O(l)$ $\Delta H = 6.01 \text{ kJ/mol}$ this is an endothermic process it requires 6.01 kJ to melt one mole of ice $H_2O(s)$ the enthalpy value will change if the number of moles varies from the 1:1 reaction stoichiometry this is an exothermic process

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web in this set of practice questions we will summarize the main concepts of thermochemistry such as the relationship between internal energy work and heat exothermic and endothermic process heat capacity constant pressure calorimetry constant volume calorimetry the enthalpy the standard enthalpies of formation and their use in

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web we will revisit this today and go on to see that if we use a special kind of thermochemical reaction called the standard enthalpy of formation we can calculate enthalpies of reactions without having to manipulate a series of individual thermochemical equations for each step

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web we can express the first law in many ways one of the more useful expressions is that the change in internal energy e of a system in any process is equal to the heat q added to the system plus the work w done on the system by its surroundings e

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