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Ions in Solution J Burgess, 1999-10-01 This outline of the principles and chemical interactions in inorganic solution chemistry delivers a course module in an area of considerable complexity Problems with solutions and tutorial hints to test comprehension have been added as a feature to check readers understanding and assist self study Exercises and projects are also provided to help readers deepen and extend their knowledge and understanding Inorganic solution chemistry is treated thoroughly Emphasis is placed upon NMR UV VIS IR Raman spectroscopy X ray diffraction and such topics as acid base behaviour stability constants and kinetics Problems and Solutions in Quantum Chemistry and Physics Charles Sidney Johnson, Lee G. Pedersen, 1986-01-01 Two hundred and eighty problems with detailed solutions plus 139 exercises all covering quantum mechanics wave mechanics angular momentum molecular spectroscopy scattering theory and related subjects An excellent problem book I would highly recommend it as a required supplement to students taking their first quantum chemistry course Journal of the American Chemical Society **Concise Chemistry of the Elements** S C Siekierski, J Burgess, 2002-10-01 The Periodic Table of the Elements is the most widely used basis for systematic discussion of inorganic chemistry Two experienced chemists encapsulate their knowledge and teaching experience in this succinct text suitable for both undergraduate and post graduate courses Part one explains how fundamental properties of atoms determine the chemical properties of elements and how and why these properties change in the Periodic Table The main properties discussed include radii and energies ionization potentials and electron affinities Particular emphasis is placed on unique properties of the first s p and d shells on the effects of filled 3d and 4d shells on the properties of p and d elements and on relativistic effects in the heavy elements The overall treatment will clarify many complex concepts Part two presents an outline of inorganic chemistry within the framework of the Periodic Table detailing the application and relevance of the principles set out in part one Explains how fundamental properties of atoms determine the chemical properties of elements and how and why these properties change in the Periodic Table The main properties discussed include radii and energies ionization potentials and electron affinities Particular emphasis is placed on unique properties of the first s p and d shells on the effects of filled 3d and 4d shells on the properties of p and d elements and on relativistic effects in the heavy elements

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Leyden, 1988 Educational Programs that Work , 1994 Resources in Education , 1995 ENC Focus , 2001

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