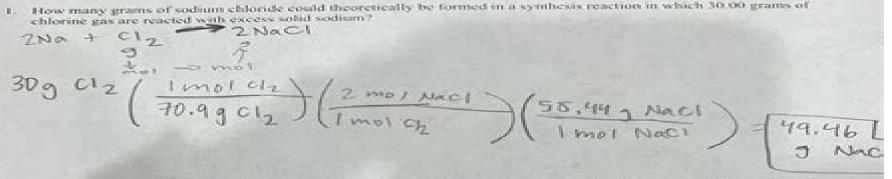
#### APChem PRACTICE WORKSHEET: Reaction Stoichiometry Review

For each of the following reaction stoichiometry problems, you must write your balanced equation, and show your work using dimensional analysis. While a map is very helpful, it is not required to be shown.

1. How many grams of sodium chloride could theoretically be formed in a synthesis reaction in which 30.00 grams of



How many molecules of water could potentially be produced by reacting excess sulfuric acid with 120, grams of sodium hydroxide in a neutralization reaction? The other product of the reaction will be sodium sulfate

$$2NaOH + H_2SO_4 \rightarrow 2H_2O + Na_2SO_4$$
 $1209$ 
 $NaOH = 40$ 



How many moles of ammonia (NH<sub>3</sub>) can theoretically be produced by the synthesis of excess hydrogen gas and 235 grams of nitrogen gas in a synthesis reaction?

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