

!!!!!! IMPORTANT NOTICE REGARDING YOUR PHONE !!!!!!

If you are replacing a defective phone, please **TURN OFF** set relocation: prior to replacing the defective phone. Failure to do so will result in loss of programming and the present assigned extension number will also be lost!

Press **FEATURE * * 266344** & Enter Password: 266344

If your Display Reads: **"A. Configuration"** Follow The Instructions Below:

Press: **SHOW**

Display will read: **"Trk/Line Data"**

Press: **NEXT**

Display will read: **"Line Access"**

Press: **NEXT** until the Display reads: **"Miscellaneous"**

Press: **SHOW**

Display will read: **"Backgrnd music"**

Press: **NEXT** until the Display reads: **"Set relocation: Y or N"**

Press the **CHANGE** softkey until it displays Y for ON / N for OFF

Press the **RLS** key.

Press **FEATURE * * 266344** & Enter Password: 266344

If your Display Reads: **"Terminals & Sets"** Follow The Instructions Below:

Press: **NEXT** until the display reads: **"System Programming"**

Press: **SHOW**

Display will read: **"Hunt Groups" or "Change DN's"**

Press: **NEXT** until the display reads: **"Featr settings"**

Press: **SHOW**

Display will read: **"Backgrnd music"**

Press: **NEXT** until the display reads: **"Set relocation"**

Press the **CHANGE** softkey to toggle the setting to Yes or No

Press the **RLS** key.

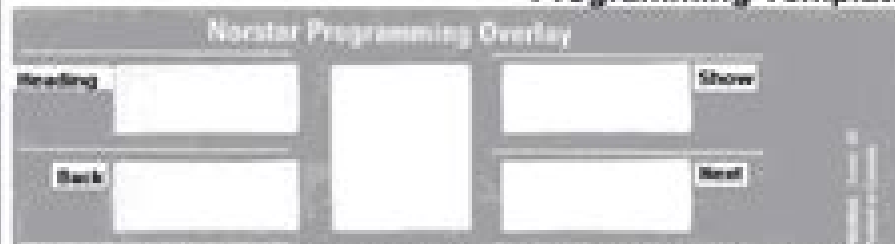


M7310



M7324

Programming Template



Place template over buttons according to your phone model

WARNING: Wait 15 min. before moving phones after making a change to Set Relocation

Rv2/6-2003

Startalk Flash User Guide

James Richard



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 answer co has 27 protons 27 electrons and 33 neutrons $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^7$
 i has 53 protons 53 electrons and 78 neutrons $1s^2 2s^2 2p^6 3s^2 3p^6 3d^{10} 4s^2 4p^6 4d^{10} 5s^2 5p^5$

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number to save room the configurations are in noble gas shorthand this means part of the electron configuration has been replaced with the element symbol of the noble gas symbol

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web elements in the second row of the periodic table place their electrons in the 2n shell as well as the 1n shell for instance lithium li has three electrons two fill the 1 s orbital and the third is placed in the 2 s orbital giving an electron configuration of 1 s 2 2 s 1

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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web the electron removed during the ionization of beryllium he 2s 2 is an s electron whereas the electron removed during the ionization of boron he 2s 2 2p 1 is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are he 2 helium 1s 2 ne 10 neon 1s 2 2s 2 2p 6

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