

States of Matter & Phase Changes

Notes

Substances can undergo a phase change into different states. Substances are in different states based on temperature and pressure. The state of a substance is determined by the relative strength of the attractive forces between its particles. The state of a substance is also determined by the relative strength of the attractive forces between its particles and the thermal energy of the particles.

More Examples

- 1. Freezing of water to ice
- 2. Condensation of steam

Learning Objectives

1. Understand the concept of phase change and the conditions for phase change.

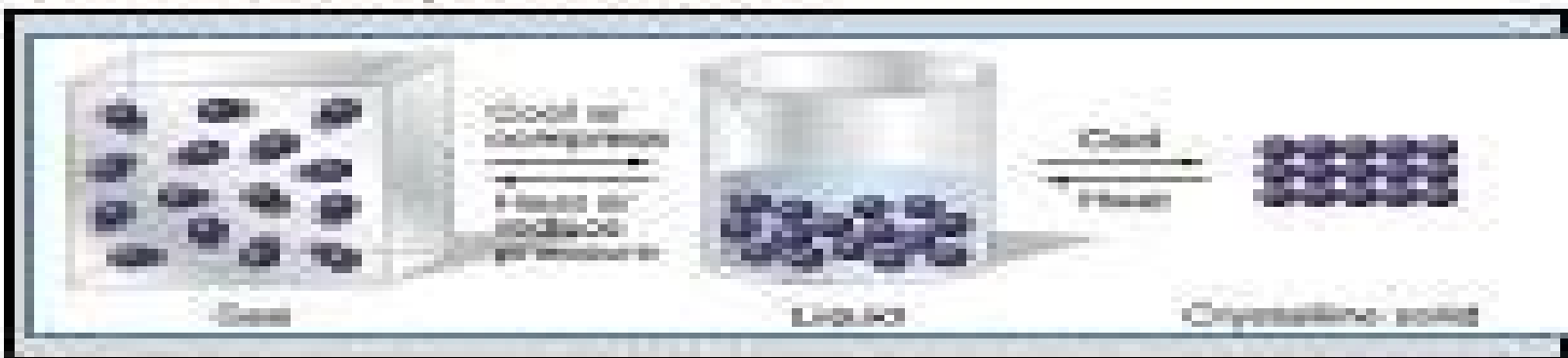
Knowledge Objectives

1. Understand the concept of phase change and the conditions for phase change.
2. Understand the concept of phase change and the conditions for phase change.
3. Understand the concept of phase change and the conditions for phase change.

Skills Objectives

1. Understand the concept of phase change and the conditions for phase change.
2. Understand the concept of phase change and the conditions for phase change.
3. Understand the concept of phase change and the conditions for phase change.

Model 1: Kinetic Theory and States of Matter



1. Understand the concept of phase change and the conditions for phase change.

State	Gas	Liquid	Solid
Shape of the substance			
Volume of the substance			
Arrangement of the particles			

States Of Matter Pogil Key

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A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the 'Oswaal Editorial Board' text.

States Of Matter Pogil Key:

Mandy Mixes It Up With States of Matter Carole Marsh, 2008 Activities for students help them explore states of matter

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