



Student Exploration: Chemical Equations Activity A & B

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. A candle is placed on one pan of a balance, and equal weight is placed on the other pan. What would happen if you lit up the candle and waited for a while?

The candle would eventually burn, thus losing mass and becoming lighter. The pan with the other object would lower as the candle burns.

2. Suppose the candle was placed in a large, sealed jar that allowed it to burn for several minutes before running out of oxygen. The candle and jar are balanced by equal weight. In this situation, what would happen if you lit up the candle and waited?

I assume the balance would stay level because the carbon dioxide is trapped within the jar, causing it to retain all of its mass despite the candle burning.

Gizmo Warm-up

Burning is an example of a **chemical reaction**. The law of **conservation of matter** states that no atoms are created or destroyed in a chemical reaction. Therefore, a balanced **chemical equation** will show the same number of each type of atom on each side of the equation.

To set up an equation in the *Chemical Equations* Gizmo, type the **chemical formulas** into the text boxes of the Gizmo. First, type in "H₂+O₂" in the **Reactants** box and "H₂O" in the **Products** box. This represents the reaction of hydrogen and oxygen gas to form water.



1. Check that the **Visual** display is chosen on each side of the Gizmo, and count the atoms.

A. How many hydrogen atoms are on the **Reactants** side?

2

Products side?

2

B. How many oxygen atoms are on the **Reactants** side?

2

Products side?

1

2. Based on what you see, is this equation currently balanced, meaning are there **equal** numbers of each reactant and product on both sides of the equation?

This equation is not currently balanced.

Activity A:

**Interpreting
chemical
formulas**

Get the Gizmo ready:

- Erase the chemical formulas in each text box.
- Check that the **Visual** displays are selected.



Student Exploration Chemical Equations Gizmo Answers

James F Hall



Student Exploration Chemical Equations Gizmo Answers:

Chemistry Equations & Answers Speedy Publishing, 2014-08-26 General chemistry inorganic chemistry organic chemistry and biochemistry are all difficult courses requiring much memorization for the student Essentially there is no easy way to learn formulas and facts This is why chemistry classes are such challenges to students even the best ones However a chemistry equations and answers study guide can help the student When used as a quick reference guide it can be used often to determine the formulas needed for various questions The astute student can cleverly devise ways to make the guide useful for test questions or other circumstances requiring one of the many chemistry equations *Chemistry Equations & Answers (Speedy Study Guide)* Speedy Publishing, 2014-11-26 Chemistry can be one of the more difficult subjects in school mainly because memorization is crucial It can be hard to memorize tons of facts and equations A chemistry equations and answers guide does help students some more than others You can refer to the study guide to determine which formula is suitable to use for the question at hand The more savvy student might consider one or more ways to creatively benefit from the guide For example creating useful tricks to aid you during a challenging test or quiz Chemistry may seem difficult but there are useful study methods

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Chemistry Equations and Answers Mark Jackson,2006-02-22 This 6 page study guide contains basic chemistry analysis and concepts designed specifically to aid science students

Chemical Equations: Chemistry Essentials Practice Workbook with 200+ Reactions to Balance Jake Mariska,2021-05-08 A chemical equation is something you will encounter every day in chemistry It s a written representation using numbers and symbols of the process that occurs during a chemical reaction Master the art of balancing chemical reactions through examples and practice 10 examples are fully solved step by step with explanations to serve as a guide Over 200 chemical equations provide ample practice Exercises start out easy and grow progressively more challenging and involved Answers to every problem are tabulated at the back of the book A chapter of pre balancing exercises helps develop essential counting skills Opening chapter reviews pertinent concepts and ideas

It's in the Equation! Law of Conservation of Mass and Chemical Equations | Grade 6-8 Physical Science Baby Professor,2024-04-15 Discover the foundational principles of chemical reactions with this insightful book perfect for middle school educators homeschooling parents and school librarians Explore the Law of Conservation of Mass and Learn about the intricacies of chemical equations highlighting their critical role in the US STEM curriculum From understanding mass conservation to mastering the art of balancing equations this book equips young learners with the essential knowledge to navigate the world of chemistry Dive into this must read for a comprehensive grasp of how substances transform in chemical reactions

The Ultimate Chemical Equations Handbook, Teacher Edition Jane D. Smith,George R. Hague,2011 *The Ultimate Chemical Equations Handbook* George R. Hague,Jane D. Smith,2001

Solutions Guide for Introductory Chemistry, Steven S. Zumdahl James F Hall,1990 **Chemical Equations** James Nixon Pendlebury,1972 *Chemistry, Problem-Solving Worktext* James E. Brady,Frederick Senese,2002-06-13 The images on the cover call attention to the relationship between macro observations and the intimate structure of chemical

substances and the changes both chemical and physical that they undergo

Fireworks

One of the ingredients is phosphorus a molecular form of which is believed to consist of linked tetrahedra of phosphorus atoms The chemical reaction of phosphorus with oxygen is partly responsible for the spectacular show of light

Carbon

The element is found in several forms including the familiar diamond and another recently discovered sooty substance that consists of soccer ball shaped molecules often referred to as buckeyballs Diamond is not the most stable form of carbon and is created from other forms of carbon at high temperatures and pressures deep within the earth

Acetylene torch

Cutting steel is possible because of the intense heat generated by the chemical reaction of acetylene with oxygen a reaction between molecules of C_2H_2 and O_2 to give CO_2 and H_2O

Hot air balloon

The air that helps it rise is heated by the combustion of molecules of propane each composed of three carbon and eight hydrogen atoms

Stormy weather

The evaporation of water serves to store energy provided by the sun Subsequent condensation of the water vapor releases this energy and is the basis of all the weather systems on our planet

Balancing Chemical Equations Learning Hub Publishing, 2020-06-22 To improve their chemistry one must perfect the skill of balancing numerous chemical equations and doing them properly This is something which you will have to practice again and again to perfect as its a key skill when it comes to high school and even in college entrance exam and some college courses In this book you will find multiple practice equations with answers in 3 terms 4 terms 5 terms and many word problems that you can work on and improve you chemistry

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