

Name: Dominick A. Credo

Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? Share

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Share 2

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? 7

2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**.

What happens? it orbits both atoms

3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**.

What happens now? it orbits both atoms

Student Exploration Covalent Bonds Gizmo Answers

Henry Sinclair Pickering

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

Student Exploration Covalent Bonds Gizmo Answers:

The Covalent Bond Henry Sinclair Pickering, 1977

Chemistry of the Covalent Bond Leallyn Burr Clapp, 1957

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Table of Contents Student Exploration Covalent Bonds Gizmo Answers

1. Understanding the eBook Student Exploration Covalent Bonds Gizmo Answers
 - The Rise of Digital Reading Student Exploration Covalent Bonds Gizmo Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Student Exploration Covalent Bonds Gizmo Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Student Exploration Covalent Bonds Gizmo Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Student Exploration Covalent Bonds Gizmo Answers
 - Personalized Recommendations

- Student Exploration Covalent Bonds Gizmo Answers User Reviews and Ratings
- Student Exploration Covalent Bonds Gizmo Answers and Bestseller Lists
- 5. Accessing Student Exploration Covalent Bonds Gizmo Answers Free and Paid eBooks
 - Student Exploration Covalent Bonds Gizmo Answers Public Domain eBooks
 - Student Exploration Covalent Bonds Gizmo Answers eBook Subscription Services
 - Student Exploration Covalent Bonds Gizmo Answers Budget-Friendly Options
- 6. Navigating Student Exploration Covalent Bonds Gizmo Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Student Exploration Covalent Bonds Gizmo Answers Compatibility with Devices
 - Student Exploration Covalent Bonds Gizmo Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Student Exploration Covalent Bonds Gizmo Answers
 - Highlighting and Note-Taking Student Exploration Covalent Bonds Gizmo Answers
 - Interactive Elements Student Exploration Covalent Bonds Gizmo Answers
- 8. Staying Engaged with Student Exploration Covalent Bonds Gizmo Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Student Exploration Covalent Bonds Gizmo Answers
- 9. Balancing eBooks and Physical Books Student Exploration Covalent Bonds Gizmo Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Student Exploration Covalent Bonds Gizmo Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Student Exploration Covalent Bonds Gizmo Answers
 - Setting Reading Goals Student Exploration Covalent Bonds Gizmo Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Student Exploration Covalent Bonds Gizmo Answers
 - Fact-Checking eBook Content of Student Exploration Covalent Bonds Gizmo Answers

- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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