

Name: Domin: ct A. Grede Date:

# Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

There are eight markers in a full set, but Flora and Frank each only have seven markers.
Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? \_\_\_\_Share\_\_\_\_

Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

share 2

#### Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the Covalent Bonds Gizmo, atoms form bonds in this way.

To begin, check that Fluorine is selected from the Select a substance menu. Click Play ( ) to see the electrons orbiting the nucleus of each atom.





- The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have?
- Click Pause (III). Drag an electron from the left atom to the right atom. Click Play.

What happens? it orbits both atoms

3. Click Pause, drag an electron from the right atom to the left, and then click Play.

What happens now? it orbits both atoms

# **Student Exploration Covalent Bonds Answer**

**VM** Jensen

#### **Student Exploration Covalent Bonds Answer:**

Strategies for Teaching Science, Levels 6-12 Barbara Houtz, 2011-06-01 Developed for grades 6 12 this rich resource provides teachers with practical strategies to enhance science instruction Strategies and model lessons are provided in each of the following overarching topics inquiry and exploration critical thinking and questioning real world applications integrating the content areas and technology and assessment Research based information and management techniques are also provided to support teachers as they implement the strategies within this resource. This resource supports core concepts Level III: Student Text hm Group, 2000-04-13 High school students whether preparing for college or of STEM instruction career need to learn greater independence and efficiency in their learning Rather than simply teach them how to study the hm Program emphasizes the need to understand their learning style empowering them to adapt to new situations throughout their lives They will learn how to improve their memory glean more from discussions prepare for exams expand their vocabulary and more Introduction to Modern Inorganic Chemistry Kenneth Malcolm Mackay, Rosemary Ann Mackay,1989 **Essentials of Genetics** William S. Klug, 2010 Key Benefit Known for its focus on problem solving conceptual understanding and practical applications this best seller is 32 pages shorter than its previous edition New features of the Seventh Edition include new Exploring Genomics exercises for selected chapters in chapter summaries that follow concept introductions for efficient review engaging case studies in each chapter an expanded Companion Website with myeBook and a new chapter on Behavioral Genetics Key Topics Introduction to Genetics Mitosis and Meiosis Mendelian Genetics Modifications of Mendelian Ratios Sex Determination and Sex Chromosomes Chromosome Mutations Variation in Number and Arrangement Linkage and Mapping in Eukaryotes Genetic Analysis and Mapping in Bacteria and Phage DNA Structure and Analysis DNA Replication and Recombination Chromosome Structure and DNA Sequence Organization The Genetic Code and Transcription Translation and Proteins Gene Mutation DNA Repair and Transposable Regulation of Genetic Expression Cancer and the Regulation of the Cell Cycle Recombinant DNA Technology and Gene Cloning Genomics and Proteomics Applications and Ethics of Genetic Engineering and Biotechnology Developmental Genetics Genetics and Behavior Quantitative Genetics Population and Evolutionary Genetics Conservation Genetics Market Intended for those interested in learning the basics of genetics **The Covalent Bond** Henry Sinclair Pickering, 1977 *Understanding* Atoms and Covalent Bonds William Eek, 2007-01-01

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In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is truly astonishing. Within the pages of "**Student Exploration Covalent Bonds Answer**," an enthralling opus penned by a highly acclaimed wordsmith, readers set about an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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