

# CHEMISTRY 101 Practice Exam 1 DR. PECK

- Directions:
- (1) Choose the best answer for each multiple choice question (numbers 1 - 15). Transfer your multiple choice answers onto the scantron.
  - (2) Do not separate any of the pages. When finished, give the proctor at the front of the room both the exam and the completed scantron. Cover your exam and your work whenever possible.

NAME \_\_\_\_\_

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\_\_\_\_\_

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(NOTE: This practice exam does not contain questions based upon topics from Chapter 5. The actual exam may or may not depending upon our ability to cover Chapter 5 in lecture before the date of the first exam.)

1. Which one of the following is a weak acid?
 

[a]  $\text{HBr}$       [b]  $\text{HClO}_4$       [c]  $\text{HNO}_3$       [d]  $\text{HCl}$       [e]  $\text{HClO}_2$
2. What is the oxidizing agent in the following reaction?
 
$$6\text{KOH}(\text{aq}) + 2\text{Cl}_2(\text{g}) \rightarrow \text{KClO}_2(\text{aq}) + 5\text{KCl}(\text{aq}) + 3\text{H}_2\text{O}(\text{l})$$

[a]  $\text{KOH}$       [b]  $\text{KCl}$       [c]  $\text{KClO}_2$       [d]  $\text{Cl}_2$  [e]  $\text{H}_2\text{O}$
3. Balance the following equation with the **smallest whole number coefficients**. What is the coefficient for  $\text{HCl}$  in the balanced equation?
 
$$\text{SnS}_3 + \text{HCl} \rightarrow \text{H}_2\text{SnCl}_6 + \text{H}_2\text{S}$$

[a] 4      [b] 6      [c] 3      [d] 12      [e] 2
4. Classify the following reaction by giving **all** of these reaction type(s) that apply.
 

I. redox                      II. combination                      III. decomposition  
 IV. displacement                      V. metathesis

$$3\text{PbO}(\text{s}) + \text{O}_2(\text{g}) \rightarrow 3\text{PbO}_2(\text{s})$$

[a] I and V    [b] only II    [c] I and II    [d] only III    [e] only V

# York College Chemistry 101 Exam 1

**Jicheng Xie**



## **York College Chemistry 101 Exam 1:**

The book delves into York College Chemistry 101 Exam 1. York College Chemistry 101 Exam 1 is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into York College Chemistry 101 Exam 1, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to York College Chemistry 101 Exam 1
- Chapter 2: Essential Elements of York College Chemistry 101 Exam 1
- Chapter 3: York College Chemistry 101 Exam 1 in Everyday Life
- Chapter 4: York College Chemistry 101 Exam 1 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of York College Chemistry 101 Exam 1. The first chapter will explore what York College Chemistry 101 Exam 1 is, why York College Chemistry 101 Exam 1 is vital, and how to effectively learn about York College Chemistry 101 Exam 1.
3. In chapter 2, the author will delve into the foundational concepts of York College Chemistry 101 Exam 1. This chapter will elucidate the essential principles that need to be understood to grasp York College Chemistry 101 Exam 1 in its entirety.
4. In chapter 3, the author will examine the practical applications of York College Chemistry 101 Exam 1 in daily life. This chapter will showcase real-world examples of how York College Chemistry 101 Exam 1 can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of York College Chemistry 101 Exam 1 in specific contexts. This chapter will explore how York College Chemistry 101 Exam 1 is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about York College Chemistry 101 Exam 1. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of York College Chemistry 101 Exam 1.

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