YEAR 9 CHEMISTRY END OF SEMESTER 1 EXAMS

Complete the table by deciding if the information indicates that the element is a metal or a non-metal.

| element | magnetic | state at room temperature | Does it conduct electricity? | metal or non-metal |
|---------|----------|------------------------------|------------------------------|--------------------------|
| A | no | liquid | yes | |
| В | yes | solid | yes | |
| C | no | liquid | no | |
| D | no | gas | no | |
| E | no | solid | yes | |

2. Look at the table. It shows the melting points and boiling points of some elements.

| element | melting point in °C | boiling point in °C | |
|---------|------------------------|------------------------|--|
| gold | 1064 | 2856 | |
| iron | 1538 | 2861 | |
| mercury | -39 | 357 | |
| oxygen | -219 | -183 | |
| sodium | 98 | 883 | |

Use the table to answer these questions.

(a) Write down the name of the element that melts at 1538°C. (1)

(b) Which element is a liquid at room temperature (20°C)? (1)

(c) Which element is a non-metal? (1)

3. Hydrogen peroxide is used to make oxygen in the laboratory.

Hydrogen peroxide breaks down to form water and oxygen.

(a) This reaction is much faster when a chemical called manganese(IV) oxide is added to the hydrogen peroxide. The manganese(IV) oxide is unchanged at the end of the reaction.

What type of chemical is manganese(IV) oxide?

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