

2. A solution of copper sulfate can be separated when heated.
Some students heated 20ml of copper sulfate solution, taken from the same batch. The diagram below shows how they set up their apparatus.



- (a) State **one** hazard associated with the reaction and the necessary safety precaution.

hazard:

safety precaution:

[2]

- (b) The students turned off their Bunsen burners and stopped heating the solution once it had begun boiling. They then left the remaining solution to evaporate for 24 hours. The students then measure the mass of the crystals collected. Below are their results.

Student	Mass of crystals (g)
Joseph	6.4
Abdullah	7.1
Susan	12.3
Candice	6.9

Calculate the **mean average** mass of crystals collected.

Do **not** include any anomalous results.

mean mass = g [2]

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Table of Contents Year 7 End Of Science Paper

1. Understanding the eBook Year 7 End Of Science Paper
 - The Rise of Digital Reading Year 7 End Of Science Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Year 7 End Of Science Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Year 7 End Of Science Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Year 7 End Of Science Paper
 - Personalized Recommendations
 - Year 7 End Of Science Paper User Reviews and Ratings
 - Year 7 End Of Science Paper and Bestseller Lists
5. Accessing Year 7 End Of Science Paper Free and Paid eBooks
 - Year 7 End Of Science Paper Public Domain eBooks
 - Year 7 End Of Science Paper eBook Subscription Services
 - Year 7 End Of Science Paper Budget-Friendly Options

6. Navigating Year 7 End Of Science Paper eBook Formats
 - ePub, PDF, MOBI, and More
 - Year 7 End Of Science Paper Compatibility with Devices
 - Year 7 End Of Science Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Year 7 End Of Science Paper
 - Highlighting and Note-Taking Year 7 End Of Science Paper
 - Interactive Elements Year 7 End Of Science Paper
8. Staying Engaged with Year 7 End Of Science Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Year 7 End Of Science Paper
9. Balancing eBooks and Physical Books Year 7 End Of Science Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Year 7 End Of Science Paper
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Year 7 End Of Science Paper
 - Setting Reading Goals Year 7 End Of Science Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Year 7 End Of Science Paper
 - Fact-Checking eBook Content of Year 7 End Of Science Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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