| 2 A.M.U. Sci./Dip. Engg. | | 2007-2008 |
|--|---|--|
| (c) 350 cm, convergent | | (c) Cl ₂ and CaSO ₄ (d) Cl ₂ and Ca(HCO ₃) ₂ |
| | 27 | The hard glass is obtained by fusing: |
| Three resistances A. B and C each of 50 and | - | A to the second of the second second black and the second second |
| connected to a battery of 10 V as about to the | | (b) a mixture of sand, lime, borax and arkan |
| | | (a) autosium carbonate and limescone |
| W | | (4) not prime carbonate lead oxide and sand |
| #v [⊥] | 28. | Which of the following metals is most reactive? |
| | | (a) Aluminium (b) lead |
| () 0 | | (c) Mercury (d) Silver Which of the following alloys contains |
| | 29. | |
| | | chromium? (a) Steel (b) Stainless steel |
| ine device used to generate electrical energy | | |
| | - | (c) Magnalium (d) Brass Which of the following is monomer of |
| | 30. | natural rubber? |
| | | (a) Chloroethene (b) Chloroprene |
| (a) 0.25 X 10 ²³ (b) 0.50 X 10 ²³ | | (c) Isoprene (d) Buta-1, 3-diene |
| | | Alkaline KMnO ₄ oxidises propanone to: |
| The value of charge / mass ratio of electron | 31. | (a) propanoic acid (b) ethanoic acid |
| | | (c) methanoic acid (d) oxalic acid |
| (a) W.K. Roenteen (b) I.I. Thomson | 22 | Which of the following metals can displace Zn |
| | 34. | from ZnSO ₄ solution? |
| | | (a) Calcium (b) Copper |
| example of combination reaction? | | (c) Iron (d) Mercury |
| (a) $Zn(s) + CuSO_{s}(sa) \rightarrow ZnSO_{s}(sa) + Cu(s)$ | 2.2 | Heating of sodium ethanoate with soda lime |
| (b) BaCl ₂ (aq) + Na ₂ SO ₂ (aq) → BaSO ₂ (s) + 2NaCl(aq) | 33. | vields: |
| (c) $CaCO_3(s) \rightarrow CaO(s) + CO_2(g)$ | | (a) ethane (b) methane |
| | | (c) ethanol (d) methanol |
| | 2.4 | The compound formed by the reaction of |
| | 34. | ethyne with bromine is: |
| | | (a) Br-CH=CH-Br (b) Br-CH ₂ -CHBr ₂ |
| The state of the s | | (c) CH ₂ =CH-Br (d) Br ₂ CH-CHBr ₂ |
| | 2.6 | Cell were first discovered by Robert Hooke in |
| | i ahah | the year: |
| | | (a) 1665 (b) 1674 (c) 1831 (d) 1835 |
| | 24 | The plant cells have a rigid cell wall that lies: |
| | 30. | (a) Outside the plasma membrane |
| (c) Li <na<k<rb (d)="" rb<k<na<li<="" td=""><td></td><td>A CAN TO THE PROPERTY OF THE P</td></na<k<rb> | | A CAN TO THE PROPERTY OF THE P |
| the state of the s | | (b) Inside the plasma membrane |
| fluorine (F) is less than: (a) Hydrogen (H) (b) Lithium (Li) | | (c) In between the plasma membranes |
| | 1900 | (d) None of the above |
| (c) Oxygen (O) (d) Chlorine (Cl) | 37. | During mitosis nucleolus and nuclea |
| Which of the following is an example of | | membrane are lost in stage: |
| strong electrolyte? | | (a) Prophase (b) Metaphase |
| (a) H ₂ CO ₁ (b) NH ₂ OH | 25.00 | (c) Anaphase (d) Telophase |
| (c) NaCl (d) COOH - COOH | 38. | An example of simple and permanent tissue is |
| | 1, 1414 | (a) Xylem (b) Phloem |
| | | (c) Selerenchyma (d) All the above |
| | 39. | The shape of squamous epithelial tissue is: |
| | | (a) Cubical (b) Flattened |
| | | (c) Pillar like (d) None of the above |
| | 40. | Fresh water sponge belongs to phylum: |
| | + | (a) Arthropoda (b) Annelida |
| (a) CaCi2 and CaCO3 (b) Ci2 and Ca(On)2 | | (o) rumenda |
| | Three resistances A, B and C each of 5Ω are connected to a battery of 10 V as shown in the figure. Calculate the current through C: SO | (c) 350 cm, convergent (d) 150 cm, divergent Three resistances A, B and C each of 5Ω are connected to a battery of 10 V as shown in the figure. Calculate the current through C: 28. (a) 2 amp (b) 1 amp 29. (c) 3 amp (d) 0.6 amp The device used to generate electrical energy is: (a) Electric motor (b) Generator (c) Galvanometer (d) Voltmeter The number of molecules in 11 g of CO ₂ are: (a) 0.25 X 10 ²⁹ (b) 0.50 X 10 ²⁹ (c) 1.00 X 10 ²⁹ (d) 1.51 X 10 ²⁹ 31. The value of charge / mass ratio of electron was determined by: (a) W K Roentgen (b) J.J.Thomson (c) Marie Curie (d) Niels Bohr Which of the following reactions is an example of combination reaction? (a) 2a(t) + Cu5O ₂ (a(t) + 2x5O ₂ (a(t) + Cu(t)) (b) 8c(3(t) + 3x5O ₂ (a(t) + BaSO ₂ (t) + 2xsC(t(a(t)) (c) CaCO ₂ (t) → CO ₂ (t) (d) C(s) + Cy(g) → CO ₂ (t) The element 'X' and 'Y' have atomic numbers 12 and 17 respectively. Element 'X' reacts with element 'Y' to form compound with molecular formula: (a) XY (b) XY ₂ (c) XY ₃ (d) Xyy (b) XY ₂ (c) XY ₃ (d) Xyy (d) Xyy, The atomic radius (pm) of Li, Na, K and Rb varies in the order: (a) Na <k<rb<1 (b)="" k<na<li="">Rb<1 (b) K<naRb (c) Li<na<k<rb (a)="" (d)="" (f)="" (kj="" affinity="" electron="" fluorine="" hyco<sub="" is="" less="" mol¹)="" of="" rb<k<na<li="" than:="" the="" value="">2 (b) NH₂OH (c) NdCH (d) COOH COOH If pH of a solution changes from 5 to 4, the change in hydrogen ion concentration shall be: (a) two times (b) five times (c) ten times (d) twenty times Bleaching powder is manufactured by the</na<k<rb></na</k<rb<1> |

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RJ Shavelson

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