



- [Index](#)
- [Language](#) •
- [General](#) •
- [Business](#) •
- [Science](#) •
- [Mathematics And Applied Science](#) •
- [Home Economics](#) •
- [Civil And Mechanical](#) •

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- [Subject Home](#)
- [1](#)
- [2](#)
- [3](#)
- [4](#)
- [5](#)

Menu

[General Comments](#)

[Weakness/Remedies](#)

[Candidate's Strength](#)

Question 1

- (a) (i) Define an acid according to the Lewis concept.
 (ii) Give one example of a Lewis acid. [3 marks]
- (b) Explain salting out in soap preparation. [2 marks]
- (c) State the reagent and condition necessary for the following conversion:

$$\text{H}-\text{C}\equiv\text{C}-\text{H} \rightarrow \text{Ag}-\text{C}\equiv\text{C}-\text{Ag}$$
 [2 marks]
- (d) What is the percentage abundance of an isotope? [2 marks]
- (e) (i) Why does the element with atomic number 18 not have an oxide?
 (ii) Explain why chlorine(I) oxide has a low melting point. [3 marks]
- (f) Describe a test to distinguish between concentrated HNO_3 and concentrated H_2SO_4 . [3 marks]
- (g) State two differences between an *electrochemical cell* and an *electrolytic cell*. [2 marks]
- (h) How does the trend in ionization energy affect the reactivity of group I element? [3 marks]
 Define the term *molecular formula*

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CH Cherryholmes



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Table of Contents Waec Answer On Chemistry Paper 2 For 2014

1. Understanding the eBook Waec Answer On Chemistry Paper 2 For 2014
 - The Rise of Digital Reading Waec Answer On Chemistry Paper 2 For 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Waec Answer On Chemistry Paper 2 For 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Waec Answer On Chemistry Paper 2 For 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Waec Answer On Chemistry Paper 2 For 2014
 - Personalized Recommendations
 - Waec Answer On Chemistry Paper 2 For 2014 User Reviews and Ratings
 - Waec Answer On Chemistry Paper 2 For 2014 and Bestseller Lists
5. Accessing Waec Answer On Chemistry Paper 2 For 2014 Free and Paid eBooks
 - Waec Answer On Chemistry Paper 2 For 2014 Public Domain eBooks
 - Waec Answer On Chemistry Paper 2 For 2014 eBook Subscription Services

- Waec Answer On Chemistry Paper 2 For 2014 Budget-Friendly Options
- 6. Navigating Waec Answer On Chemistry Paper 2 For 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Waec Answer On Chemistry Paper 2 For 2014 Compatibility with Devices
 - Waec Answer On Chemistry Paper 2 For 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Waec Answer On Chemistry Paper 2 For 2014
 - Highlighting and Note-Taking Waec Answer On Chemistry Paper 2 For 2014
 - Interactive Elements Waec Answer On Chemistry Paper 2 For 2014
- 8. Staying Engaged with Waec Answer On Chemistry Paper 2 For 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Waec Answer On Chemistry Paper 2 For 2014
- 9. Balancing eBooks and Physical Books Waec Answer On Chemistry Paper 2 For 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Waec Answer On Chemistry Paper 2 For 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Waec Answer On Chemistry Paper 2 For 2014
 - Setting Reading Goals Waec Answer On Chemistry Paper 2 For 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Waec Answer On Chemistry Paper 2 For 2014
 - Fact-Checking eBook Content of Waec Answer On Chemistry Paper 2 For 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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