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Chemistry Paper 2 (Practical), WASSCE (SC), 2023

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Question 1

(a) What is transition element? [2 marks]

(b) Consider the electron configuration of the following elements:
 A = 2: 8:6; B = 2: 8:2; C = 2:8:1; D = 2:8:8

State the element which forms a:

- (i) doubly charged cation;
 - (ii) soluble trioxocarbonate (IV).
- [2 marks]

(c) Explain **briefly** why there is a general increase in the **first** ionization energies of the elements across the period in the periodic table.
 [2 marks]

(d) Give two examples of an aliphatic compound.
 [2 marks]

(e) Explain **briefly** why alkanols are stronger bases than water
 [3 marks]

(f) State the **major** raw materials used in the solvay process.
 [3 marks]

(g) What is geometric isomerism?
 [2 marks]

(h) Give a reason why water gas is a better fuel than producer gas.

(i) Define the term heat of combustion.

(j) (i) State Faraday second law of electrolysis.

(ii) Calculate the amount of silver deposited when 10920 coulombs of electricity is passed through a solution of a silver salt.

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AW Rasmussen



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