

QUESTION 4

(a) (i) Dalton's law of partial pressures states that if there is a mixture of gases which do not react chemically together, then the total pressure exerted by the mixture is the sum of the partial pressures of the individual gases that make up the mixture.

$$\begin{aligned}\text{(ii) Partial pressure of O}_2 &= \left(\frac{2}{2+3+5} \right) \times 210 \\ &= \frac{2}{10} \times 210 \\ &= 42 \text{ Nm}^{-2}\end{aligned}$$

$$\begin{aligned}\text{Partial pressure of N}_2 &= \left(\frac{5}{2+3+5} \right) \times 210 \\ &= \frac{5}{10} \times 210 \\ &= 105 \text{ Nm}^{-2}\end{aligned}$$

(b) (i) The concentration of a solution is the amount of solute in 1 dm^3 of the solution, expressed in mol/dm^3 or g/dm^3 .

$$\begin{aligned}\text{(ii) mass concentration} &= \text{molarity} \times \text{molar mass} \\ &= 0.10 \times 106 \\ &= 10.6 \text{ g/dm}^3\end{aligned}$$

(iii) Weigh 10.6 g of Na_2CO_3 accurately

- Dissolve it in a small amount of water in a volumetric flask
- Add water to make up to 1 dm^3 mark.

(c) (i) Chlorine:

- It is a greenish-yellow gas
- It is about 2.5 times denser than air

hydrogen chloride:

- Pure hydrogen chloride is a colourless gas with irritating smell
- It turns damp litmus paper red.

Waec May June 2014 For Chemistry Obj And Essay Question

Liying Dong



Waec May June 2014 For Chemistry Obj And Essay Question:

This book delves into Waec May June 2014 For Chemistry Obj And Essay Question. Waec May June 2014 For Chemistry Obj And Essay Question is a crucial topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Waec May June 2014 For Chemistry Obj And Essay Question, encompassing both the fundamentals and more intricate discussions.

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Table of Contents Waec May June 2014 For Chemistry Obj And Essay Question

1. Understanding the eBook Waec May June 2014 For Chemistry Obj And Essay Question
 - The Rise of Digital Reading Waec May June 2014 For Chemistry Obj And Essay Question
 - Advantages of eBooks Over Traditional Books
2. Identifying Waec May June 2014 For Chemistry Obj And Essay Question
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Waec May June 2014 For Chemistry Obj And Essay Question
 - User-Friendly Interface
4. Exploring eBook Recommendations from Waec May June 2014 For Chemistry Obj And Essay Question
 - Personalized Recommendations
 - Waec May June 2014 For Chemistry Obj And Essay Question User Reviews and Ratings
 - Waec May June 2014 For Chemistry Obj And Essay Question and Bestseller Lists
5. Accessing Waec May June 2014 For Chemistry Obj And Essay Question Free and Paid eBooks
 - Waec May June 2014 For Chemistry Obj And Essay Question Public Domain eBooks
 - Waec May June 2014 For Chemistry Obj And Essay Question eBook Subscription Services
 - Waec May June 2014 For Chemistry Obj And Essay Question Budget-Friendly Options
6. Navigating Waec May June 2014 For Chemistry Obj And Essay Question eBook Formats
 - ePub, PDF, MOBI, and More
 - Waec May June 2014 For Chemistry Obj And Essay Question Compatibility with Devices
 - Waec May June 2014 For Chemistry Obj And Essay Question Enhanced eBook Features

7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Waec May June 2014 For Chemistry Obj And Essay Question
 - Highlighting and Note-Taking Waec May June 2014 For Chemistry Obj And Essay Question
 - Interactive Elements Waec May June 2014 For Chemistry Obj And Essay Question
8. Staying Engaged with Waec May June 2014 For Chemistry Obj And Essay Question
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Waec May June 2014 For Chemistry Obj And Essay Question
9. Balancing eBooks and Physical Books Waec May June 2014 For Chemistry Obj And Essay Question
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Waec May June 2014 For Chemistry Obj And Essay Question
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Waec May June 2014 For Chemistry Obj And Essay Question
 - Setting Reading Goals Waec May June 2014 For Chemistry Obj And Essay Question
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Waec May June 2014 For Chemistry Obj And Essay Question
 - Fact-Checking eBook Content of Waec May June 2014 For Chemistry Obj And Essay Question
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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