Standard Heats of Formation

Some Heats of Formation, ΔH_f° (kJ/mol)

CO(g)	-111	$C_2H_2(g)$	227	Ag+(aq)	106
$CO_2(g)$	-394	$C_2H_4(g)$	52	Na ⁺ (aq)	-240
$H_2O(l)$	-286	$C_2H_6(g)$	-85	NO ₃ -(aq)	-207
$NH_3(g)$	-46	CH ₃ OH(g)	-201	Cl ⁻ (aq)	-167
$N_2H_4(g)$	95.4	$C_2H_5OH(g)$	-235	AgCl(s)	-127
HCl(g)	-92	$C_6H_6(l)$	49	Na ₂ CO ₃ (s)	-1131

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