

[Atomic Structure Worksheet]

- 1 How many protons and neutrons are present in an atom of calcium (Mass number = 40)?

Protons = _____

Neutrons = _____

- 2 How many protons and neutrons are present in an atom of krypton (Mass number = 84)?

Protons = _____

Neutrons = _____

- 3 What is the name, symbol, and charge of an ion that contains 12 protons and 10 electrons?

Name = _____

Symbol = _____

Charge = _____

- 4 What is the name, symbol, and charge of an ion that contains 15 protons and 18 electrons?

Name = _____

Symbol = _____

Charge = _____

- 5 A neutral atom has 85 protons and 125 neutrons. Show calculations wherever necessary.

a What is its atomic number? _____

b What is its mass number? _____

c How many electrons does the atom have? _____

d What is the name of the atom? _____

- 6 Assuming all the atoms have neutral charge, complete the following table using your knowledge on atomic structure.

	Protons	Electrons	Neutrons	Mass number
Atom 1		8	8	
Atom 2			14	28
Atom 3	22			48
Atom 4		47		107
Atom 5		83	126	
Atom 6	87			223

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 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Standarized Test Prep Atomic Structure Pearson Chemistry
 - User-Friendly Interface
4. Exploring eBook Recommendations from Standarized Test Prep Atomic Structure Pearson Chemistry
 - Personalized Recommendations
 - Standarized Test Prep Atomic Structure Pearson Chemistry User Reviews and Ratings

- Standardized Test Prep Atomic Structure Pearson Chemistry and Bestseller Lists
- 5. Accessing Standardized Test Prep Atomic Structure Pearson Chemistry Free and Paid eBooks
 - Standardized Test Prep Atomic Structure Pearson Chemistry Public Domain eBooks
 - Standardized Test Prep Atomic Structure Pearson Chemistry eBook Subscription Services
 - Standardized Test Prep Atomic Structure Pearson Chemistry Budget-Friendly Options
- 6. Navigating Standardized Test Prep Atomic Structure Pearson Chemistry eBook Formats
 - ePub, PDF, MOBI, and More
 - Standardized Test Prep Atomic Structure Pearson Chemistry Compatibility with Devices
 - Standardized Test Prep Atomic Structure Pearson Chemistry Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Standardized Test Prep Atomic Structure Pearson Chemistry
 - Highlighting and Note-Taking Standardized Test Prep Atomic Structure Pearson Chemistry
 - Interactive Elements Standardized Test Prep Atomic Structure Pearson Chemistry
- 8. Staying Engaged with Standardized Test Prep Atomic Structure Pearson Chemistry
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Standardized Test Prep Atomic Structure Pearson Chemistry
- 9. Balancing eBooks and Physical Books Standardized Test Prep Atomic Structure Pearson Chemistry
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Standardized Test Prep Atomic Structure Pearson Chemistry
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Standardized Test Prep Atomic Structure Pearson Chemistry
 - Setting Reading Goals Standardized Test Prep Atomic Structure Pearson Chemistry
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Standardized Test Prep Atomic Structure Pearson Chemistry
 - Fact-Checking eBook Content of Standardized Test Prep Atomic Structure Pearson Chemistry
 - Distinguishing Credible Sources

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 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
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 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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