

Physical and Chemical Changes and Properties of Matter Worksheet

Classify the following as chemical change (cc), chemical property (cp), physical change (pc), or physical property (pp).

- | | |
|-------------------------------------|-------------------------------|
| 1. <u>PP</u> Heat conductivity | 8. <u>CP</u> Combustible |
| 2. <u>CC</u> Silver tarnishing | 9. <u>PC</u> Water freezing |
| 3. <u>PC</u> sublimation | 10. <u>CC</u> Wood burning |
| 4. <u>PC</u> magnetizing steel | 11. <u>CP</u> Acid resistance |
| 5. <u>PP</u> length of metal object | 12. <u>PP</u> Brittleness |
| 6. <u>PC</u> shortening melting | 13. <u>CC</u> Milk souring |
| 7. <u>CC</u> exploding dynamite | 14. <u>CC</u> baking bread |

Identify the following as being true or false to the left of the sentence.

- T 15. A change in size or shape is a physical change.
T 16. A chemical change means a new substance with new properties was formed.
F 17. An example of a chemical change is when water freezes.
T 18. When platinum is heated, then cooled to its original state, we say this is a physical change.
F 19. When milk turns sour, this is a physical change because a change in odor does not indicate a chemical change.
T 20. When citric acid and baking soda mix, carbon dioxide is produced and the temperature decreases. This must be a chemical change.

Identify each of the following as a physical or chemical change.

21. CC You leave your bicycle out in the rain and it rusts.
22. PC A sugar cube dissolves.
23. CC Scientist break-up water into oxygen and hydrogen gas.
24. CC Burning coal for a barbecue.
25. PC Trimming a bush because it has grown too tall.

Classifying Matter Worksheet

Classify each of the following substances as an element, a compound, a solution (homogeneous mixture), or a heterogeneous mixture.

- | | | |
|----------------------------------|---|-----------------------------------|
| 1. Sand
heterogeneous mixture | 2. Salt
Compound | 3. Pure Water
compound |
| 4. Soil
heterogeneous mixture | 5. Soda just opened
homogenous mixture | 6. Pure air
homogenous mixture |

Unit 1 Physical And Chemical Changes Answers

Eugene Gustave Eberle



Unit 1 Physical And Chemical Changes Answers:

Chemistry-vol-I Dr Vijay Sarda, Dr A C Handa, Dr K K Arora, A text book on Chemistry Ohio State University Bulletin, 1913 Science - Earth and Space Life - Physical, 2003-09 Provides background information for the teacher cross curricular activities projects and experiments and assessments p 4 of cover **The Ohio State University Bulletin** Ohio State University, 1913 **Biochemistry and Analytical Techniques** Mr. Rohit Manglik, 2024-05-16 Introduces biochemical processes and the lab techniques used to analyze body fluids and functions Supports diagnostic and clinical decision making *OHIO STATE UNIVERSITY BULLETIN: ENTRANCE EXAMINATION QUESTIONS* Ohio State University, 1913 Food Fundamentals and Chemistry Mr. Rohit Manglik, 2024-03-06 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels Laboratory Chemistry for Girls Agnes French Jaques, 1923 **Chemistry in the Community (ChemCom)** American Chemical Society, 2011-06-17 Touted as the most successful NSF funded project published Chemistry in the Community ChemCom by the American Chemical Society ACS offers a meaningful and memorable chemistry program for all levels of high school students ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios Centered on decision making activities where students are responsible for generating data in an investigating analyzing that data and then applying their chemistry knowledge to solve the presented problem The text is intensively laboratory based with all 39 of the investigations integrated within the text not separate from the reading With the ChemCom program students learn more organic and biochemistry more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks all within the relevance of solving problems that arise in everyday life Meticulously updated to meet the needs of today s teachers and students the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards Incorporating advances in learning and cognitive sciences ChemCom s wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards Correlations are available showing how closely aligned ChemCom is to these and other state standards Harcourt Science: Physical science, [grade] 3, units E and F, teacher's ed, 2000 The Student in Pharmacy Eugene Gustave Eberle, 1913 **Harcourt Science: Physical science, [grade] 5, Units E and F, teacher's ed**, 2000 **Chemistry in Everyday Life** Frederic Barclay Emery, Elzy F. Downey, Roscoe Edward Davis, Charles E. Boynton, 1923 Northwest Journal of Education, 1898 Inorganic Chemistry Hamilton Perkins Cady, 1912 **Science** Richard Moyer, 2000 For grades 1 6 **Chemical Pathology** Harry Gideon Wells, 1925 **Modern Physical Science** George R. Tracy, 1983 Loose-Leaf Version for Chemical Principles Peter Atkins, Loretta Jones, Leroy Laverman, 2016-01-07 Written for calculus inclusive general chemistry courses Chemical Principles helps students develop

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