

Chemical Equilibrium - Assignment 1

Questions

1. A chemical reaction is at equilibrium when
 - (a) Reactants are completely transformed into products
 - (b) The rates of forward and backward reactions are equal
 - (c) Formation of products is minimized
 - (d) Equal amounts of reactants and products are present

2. Which of the following reactions can reach equilibrium in open vessel?
 - (a) $\text{NH}_4\text{HS(s)} \rightleftharpoons \text{NH}_3\text{(g)} + \text{H}_2\text{S(g)}$
 - (b) $\text{N}_2\text{(g)} + 3\text{H}_2\text{(g)} \rightleftharpoons 2\text{NH}_3\text{(g)}$
 - (c) $\text{BaCl}_2\text{(s)} + \text{H}_2\text{SO}_4\text{(aq)} \rightleftharpoons \text{BaSO}_4\text{(s)} + 2\text{HCl(aq)}$
 - (d) All of these

3. For the reaction, $\text{CO(g)} + \text{Cl}_2\text{(g)} \rightleftharpoons \text{COCl}_2\text{(g)}$ the $\frac{K_p}{K_c}$ is equal to
 - (a) $\sqrt{RT}$
 - (b) RT
 - (c) $\frac{1}{RT}$
 - (d) 1.0

4. The equilibrium constant K_c for the following reaction at 500 K is, $\text{A}_2\text{(g)} + \text{B}_2\text{(g)} \rightleftharpoons 2\text{AB}$; $K_c = 60$
 Select the correct statement/statements:
 - (a) Rate constant for forward reaction k_f = Rate constant for backward reaction k_b
 - (b) Rate constant for forward reaction $k_f >$ Rate constant for backward reaction k_b
 - (c) Rate constant for forward reaction $k_f <$ Rate constant for backward reaction k_b
 - (d) Value of equilibrium constant changes with the use of catalyst.

5. What is observed at equilibrium point in the reaction?
 $\text{PCl}_{5\text{(g)}} \rightleftharpoons \text{PCl}_{3\text{(g)}} + \text{Cl}_{2\text{(g)}}$
 - (a) Equal volumes of PCl_5 , PCl_3 and Cl_2 are present.
 - (b) Equal masses of PCl_5 , PCl_3 and Cl_2 are present.
 - (c) The concentrations of PCl_5 , PCl_3 and Cl_2 become constant.
 - (d) Reaction comes to a stop.

6. For the reaction: $\text{H}_2\text{(g)} + \text{I}_2\text{(g)} \rightleftharpoons 2\text{HI(g)}$, the equilibrium constant K_c changes with

Unit 3 Chemical Equilibrium Assignment 2 Answers

RC Schank



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