

Unit 2 Worksheet: St. Michael's Concentration

Part A: Dissolving Metal Concentration

1. Cadmium chloride has a solubility in water for pressures at 1.000 mol/L. Explain what this concentration tells you about the CdCl_2 solution. What might a chemical use this method?

2. A 20.00 g sample of potassium nitrate is dissolved in enough water for exactly 200.0 mL of solution. Calculate the exact concentration of this potassium nitrate.

3. Find the molarity of a solution made from 2.50 g of CaCl_2 dissolved in enough water for exactly 1.00 L.

4. An unknown metal nitrate solution is "titrated" and reduced to prepare a 10% solution of 5.00 g of metal compound in enough water for exactly 100.0 mL. Calculate the molarity of metal in this solution.

5. What metal compound might be involved in a 10% solution of 10.00 g NaCl solution from 10.0 g of NaCl x 10?

Unit 9 Worksheet 3 Molar Concentration Stoichiometry

Bowker Editorial Staff



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Cambridge IGCSETM Chemistry Teacher's Guide (Collins Cambridge IGCSETM) Chris Sunley, 2022-02-03

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the 1993 edition this iteration does not assume that students lab technicians and scientists have mastered the prerequisite calculation skills for quantitative problems in the chemical biomedical sciences A new chapter focuses on using spreadsheets and laboratory information management systems Other chapters cover calculations and techniques relevant to reagents chemical reactions properties of gases and solutions pH and buffer preparation spectrophotometry enzyme assays and radioactivity Also included are derivations of some key equations quick reference guides and an index to the practical examples Efiok is with the National Heart Lung and Blood Institute National Institutes of Health Eduok is in the chemistry department at Xavier U of Louisiana c Book News Inc **R.R. Bowker's Software for Schools**, 1987 *The Software Encyclopedia 2000* Bowker Editorial Staff, 2000-05

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