

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? *Frank can share his red marker with Flora and Flora can share her blue marker with Frank.*
2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set? *Olivia can share her purple and green markers with Otto and Otto can share his black and brown markers with Olivia.*

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.



To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.

1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? *Each fluorine atom has seven valence electrons.*
2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**. What happens? *The electrons that was dragged now orbits both atoms.*
3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**. What happens now? *A pair of electrons now orbit both atoms.*

Student Exploration Covalent Bond Gizmo Answer Key

Clemens Wendtner

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Student Exploration Covalent Bond Gizmo Answer Key

1. Understanding the eBook Student Exploration Covalent Bond Gizmo Answer Key
 - The Rise of Digital Reading Student Exploration Covalent Bond Gizmo Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Student Exploration Covalent Bond Gizmo Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Student Exploration Covalent Bond Gizmo Answer Key
 - User-Friendly Interface

4. Exploring eBook Recommendations from Student Exploration Covalent Bond Gizmo Answer Key
 - Personalized Recommendations
 - Student Exploration Covalent Bond Gizmo Answer Key User Reviews and Ratings
 - Student Exploration Covalent Bond Gizmo Answer Key and Bestseller Lists
5. Accessing Student Exploration Covalent Bond Gizmo Answer Key Free and Paid eBooks
 - Student Exploration Covalent Bond Gizmo Answer Key Public Domain eBooks
 - Student Exploration Covalent Bond Gizmo Answer Key eBook Subscription Services
 - Student Exploration Covalent Bond Gizmo Answer Key Budget-Friendly Options
6. Navigating Student Exploration Covalent Bond Gizmo Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Student Exploration Covalent Bond Gizmo Answer Key Compatibility with Devices
 - Student Exploration Covalent Bond Gizmo Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Student Exploration Covalent Bond Gizmo Answer Key
 - Highlighting and Note-Taking Student Exploration Covalent Bond Gizmo Answer Key
 - Interactive Elements Student Exploration Covalent Bond Gizmo Answer Key
8. Staying Engaged with Student Exploration Covalent Bond Gizmo Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Student Exploration Covalent Bond Gizmo Answer Key
9. Balancing eBooks and Physical Books Student Exploration Covalent Bond Gizmo Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Student Exploration Covalent Bond Gizmo Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Student Exploration Covalent Bond Gizmo Answer Key
 - Setting Reading Goals Student Exploration Covalent Bond Gizmo Answer Key
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Student Exploration Covalent Bond Gizmo Answer Key
 - Fact-Checking eBook Content of Student Exploration Covalent Bond Gizmo Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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