

Student Exploration: Calorimetry Lab

Vocabulary: calorie, calorimeter, joule, specific heat capacity

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. The Latin word *calor* means "heat," and *meter* comes from the Greek word meaning "to measure." What do you think a **calorimeter** does? **The calorimeter is an instrument used to measure heat or heat energy and any changes in it.**
2. Where have you heard the word **calorie** before? What do you think a calorie is? **When I hear calorie I think it's about losing weight with your foods. I think a calorie is an ingredient, or type of energy. I also know that people sometimes tend to worry about how much calories they intake, and whatnot. And I also know that you can burn calories, which can suggest that it is a type of energy.**

Gizmo Warm-up

A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up.

Calorimeters can be used to find a substance's **specific heat capacity**. You will use the *Calorimetry Lab Gizmo* to determine the specific heat capacities of various substances.



1. On the SIMULATION pane, select **Copper**. Use the slider to set its **Mass** to 200 g. Set the **Water mass** to 200 g. Check that the **Water temp** is set to 30.0 °C and the copper's **Temp** is 90 °C. Select the GRAPH tab, and click **Play** ().

A. What was the **Final temperature** of the copper and the water? **34.96°C**

B. How much did the temperature of the copper change? **55.04**

C. How much did the temperature of the water change? **4.96**

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BM King

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the 'BM King' text.

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