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Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? Share

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Share 2

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? 7

2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**.

What happens? it orbits both atoms

3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**.

What happens now? it orbits both atoms

Student Exploration Covalent Bonds All Answers

Marcia A. Mardis



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Exploration of the Structure of Atom Chandan Sengupta, First Publication April 2019 Revised Publication October 2022 Second Revised Edition July 2023 Third Revised Edition December 2024 Published by Chandan Senguta Printed by IECIT Printing and ublication Services Resource Centre Arabinda Nagar Bankura 722101 W B India Topics Covered Atoms and Molecules Structure of Atom Bonding Mechanism and Chemical Reactions Mechanism of Bonding This book is suitable for students of Class 9 to Class 11 Students aspiring for Pre Medical Entrance Examination can also get adequate support Additional Hard Copies can also be obtained from Chandan Sukumar Sengupta Arabinda Nagar Bankura 722101 WB Write to Us for more materials Oswaal CBSE & NCERT One for All Class 10 Science | With Topic Wise Notes For 2025 Board Exam Oswaal Editorial Board,2024-05-21 Description of the product 1 NCERT Textbook Exemplar for Concepts Recall 2 Previous Years Questions for Exam Trends Insights 3 Competency Based Questions for Holistic Skill Development 4 NEP Compliance with Artificial Intelligence Art Integration *Strategies for Teaching Science, Levels 6-12* Barbara Houtz,2011-06-01 Developed for grades 6 12 this rich resource provides teachers with practical strategies to enhance science instruction Strategies and model lessons are provided in each of the following overarching topics inquiry and exploration critical thinking and questioning real world applications integrating the content areas and technology and assessment Research based information and management techniques are also provided to support teachers as they implement the strategies within this resource This resource supports core concepts of STEM instruction *Solid State Physics* Philip Hofmann,2022-05-25 Solid State Physics Enables readers to easily understand the basics of solid state physics Solid State Physics is a successful short textbook that gives a clear and concise introduction to its subject The presentation is suitable for students who are exposed to this topic for the first time Each chapter starts with basic principles and gently progresses to more advanced concepts using easy to follow explanations and keeping mathematical formalism to a minimum This new edition is thoroughly revised with easier to understand descriptions of metallic and covalent bonding a straightforward proof of Bloch s theorem a simpler approach to the nearly free electron model and enhanced pedagogical features such as more than 100 discussion questions 70 problems including problems to train the students skills to find computational solutions and multiple choice questions at the end of each chapter with solutions in the book for self training Solid State Physics introduces the readers to Crystal structures and underlying bonding mechanisms The mechanical and vibrational properties of solids Electronic properties in both a classical and a quantum mechanical picture with a treatment of the electronic phenomena in metals semiconductors and insulators More advanced subjects such as magnetism superconductivity and phenomena emerging for nano scaled solids For bachelor s students in physics materials sciences engineering sciences and chemistry Solid State Physics serves as an introductory textbook with many helpful supplementary learning resources included throughout the text and available online to aid in reader comprehension **Level III: Student**

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