



Acid-Base Equilibria

14.3 – Relative Strengths of Acids and Bases

		ACID		BASE	
Strong acids		HCl	↑ Base strength increasing ↓ ↑ Acid strength increasing ↓	Cl ⁻	negligible bases ^{SA → neg}
		H ₂ SO ₄		HSO ₄ ⁻	
		HNO ₃		NO ₃ ⁻	
		H ₃ O ⁺ (aq)		H ₂ O	
Weak acids		HSO ₄ ⁻		SO ₄ ²⁻	WA → WB WB → WA
		H ₃ PO ₄		H ₂ PO ₄ ⁻	
		HF		F ⁻	
		CH ₃ COOH		CH ₃ COO ⁻	
		H ₂ CO ₃		HCO ₃ ⁻	
		H ₂ S		HS ⁻	
		H ₂ PO ₄ ⁻		HPO ₄ ²⁻	
		NH ₄ ⁺		NH ₃	
		HCO ₃ ⁻		CO ₃ ²⁻	
		HPO ₄ ²⁻		PO ₄ ³⁻	
Negligible acidity		H ₂ O		OH ⁻	strong base ^{SB → neg}
		OH ⁻		O ²⁻	
		H ₂		H ⁺	
		CH ₄		CH ₃ ⁻	

Strengths Of Acids And Bases 19 3 Key

Brian W. Pfennig



Strengths Of Acids And Bases 19 3 Key:

Acids and Bases Brian G. Cox, 2013-01-31 This book seeks to enhance our understanding of acids and bases by reviewing and analysing their behaviour in non aqueous solvents The behaviour is related where possible to that in water but correlations and contrasts between solvents are also presented

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linear teaching programmes in areas of basic organic chemistry was developed for use by first year undergraduates at Sussex University where it has proved extremely popular The concepts treated in the seven programmes are fundamental to any introductory course in organic chemistry and include certain intellectual techniques such as curly arrows which students often find difficult but which need to be fully mastered With tutorial time increasingly short this self study text should prove invaluable to students of chemistry biology or pre clinical medicine taking a first course in organic chemistry

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Shriver and Atkins' Inorganic Chemistry Peter Atkins, 2010 Inorganic Chemistry fifth edition represents an integral part of a student's chemistry education Basic chemical principles are set out clearly in Foundations and are fully developed throughout the text culminating in the cutting edge research topics of the Frontiers which illustrate the dynamic nature of inorganic chemistry

Physical Science Carson-Dellosa Publishing, 2015-03-09 Physical Science for grades 5 to 12 is designed to aid in the review and practice of physical science topics Physical Science covers topics such as scientific measurement force and energy matter atoms and elements magnetism and electricity The book includes realistic diagrams and engaging activities to support practice in all areas of physical science The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science

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Basic

Inorganic Chemistry F. Albert Cotton, Geoffrey Wilkinson, Paul L. Gaus, 1995-01-18 Explains the basics of inorganic chemistry with a primary emphasis on facts then uses the student's growing factual knowledge as a foundation for discussing the important principles of periodicity in structure bonding and reactivity New to this updated edition improved treatment of atomic orbitals and properties such as electronegativity novel approaches to the depiction of ionic structures nomenclature for transition metal compounds quantitative approaches to acid base chemistry Wade's rules for boranes and carboranes the chemistry of major new classes of substances including fullerenes and silenes plus a chapter on the inorganic solid state

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chapters of texts giving it only a cursory overview Covers atomic and molecular term symbols symmetry coordinates in vibrational spectroscopy using the projection operator method polyatomic MO theory band theory and Tanabe Sugano diagrams Includes a heavy dose of group theory in the primary inorganic textbook most of the pedagogical benefits of integration and reinforcement of this material in the treatment of other topics such as frontier MO acid base theory band theory of solids inorganic photochemistry the Jahn Teller effect and Wade s rules are fully realized Very physical in nature compare to other textbooks in the field taking the time to go through mathematical derivations and to compare and contrast different theories of bonding in order to allow for a more rigorous treatment of their application to molecular structure bonding and spectroscopy Informal and engaging writing style worked examples throughout the text unanswered problems in every chapter contains a generous use of informative colorful illustrations

Karp's Cell and Molecular Biology Gerald Karp, Janet Iwasa, Wallace Marshall, 2020-02-19 Karp s Cell and Molecular Biology delivers a concise and illustrative narrative that helps students connect key concepts and experimentation so they better understand how we know what we know in the world of cell biology This classic text explores core concepts in considerable depth often adding experimental detail It is written in an inviting style and at mid length to assist students in managing the plethora of details encountered in the Cell Biology course The 9th Edition includes two new sections and associated assessment in each chapter that show the relevance of key cell biology concepts to plant cell biology and bioengineering

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