



ELECTRONS IN ATOMS

Chapter Test B

A. Matching

Match each item with the correct definition or activity. Write the letter of the correct definition or activity in the space provided.

Column A

- _____ 1. A model of an atom in which electrons are arranged in concentric shells.
- _____ 2. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical.
- _____ 3. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.
- _____ 4. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.
- _____ 5. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.
- _____ 6. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.
- _____ 7. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.
- _____ 8. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.
- _____ 9. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.
- _____ 10. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.

Column B

- a. Bohr's model of the atom
- b. Rutherford's model of the atom
- c. Thomson's model of the atom
- d. Dalton's model of the atom
- e. Schrödinger's model of the atom
- f. Heisenberg's model of the atom
- g. de Broglie's model of the atom
- h. Planck's model of the atom
- i. Einstein's model of the atom
- j. Bohr's model of the atom

B. Multiple Choice

Choose the best answer for each question.

- _____ 11. Which of the following is the best description of the Bohr model?
a. A model of an atom in which electrons are arranged in concentric shells.
b. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical.
c. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.
d. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.
- _____ 12. Which of the following is the best description of the Bohr model?
a. A model of an atom in which electrons are arranged in concentric shells.
b. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical.
c. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.
d. A model of an atom in which electrons are arranged in concentric shells, but the shells are not perfectly spherical and the electrons are not arranged in concentric shells.

Way In Which Electrons Are Arranged Chapter 5 Test

Melvin D. Joesten, James L. Wood



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