

1. (5 points) Table 4.8 (Snoeyink and Jenkins) shows the concentration of carbonate species for solutions with distilled water and 10^{-5}M solutions of H_2CO_3^* ; sodium bicarbonate, NaHCO_3 ; and sodium carbonate, Na_2CO_3 , in a system open to the atmosphere and in a closed system.
- Why does the concentration of H_2CO_3^* remain constant in the open system (across the different solutions) and decreases in a closed system?
 - For both systems, the concentration of H^+ decreases when comparing a solution of distilled water and 10^{-5}M solutions of H_2CO_3^* ; NaHCO_3 ; Na_2CO_3 with lowest concentration observed for Na_2CO_3 . The decrease in H^+ concentration is greater for a closed system than for an open system. Explain this difference.

TABLE 4-8 Comparison of Composition and pH of Open and Closed Carbonate Systems

Solution	Closed System (Fig. 4-17)						
	$[\text{H}^+]$	$[\text{OH}^-]$	$[\text{H}_2\text{CO}_3^*]$	$[\text{HCO}_3^-]$	$[\text{CO}_3^{2-}]$	pH	$C_{\text{T,CO}_3}$
$10^{-5}\text{ M H}_2\text{CO}_3^*$	$10^{-6.3}$	$10^{-7.7}$	$10^{-5.1}$	$10^{-6.7}$	$10^{-10.2}$	5.7	$10^{-5.1}$
10^{-5} M NaHCO_3	$10^{-7.8}$	$10^{-6.2}$	$10^{-6.4}$	10^{-5}	$10^{-10.4}$	7.6	$10^{-5.1}$
$10^{-5}\text{ M Na}_2\text{CO}_3$	10^{-9}	10^{-5}	$10^{-7.6}$	10^{-5}	$10^{-4.2}$	9.0	$10^{-5.1}$
Solution	Open System						
	$[\text{H}^+]$	$[\text{OH}^-]$	$[\text{H}_2\text{CO}_3^*]$	$[\text{HCO}_3^-]$	$[\text{CO}_3^{2-}]$	pH	$C_{\text{T,CO}_3}$
$10^{-5}\text{ M H}_2\text{CO}_3^*$	$10^{-6.3}$	$10^{-7.7}$	10^{-5}	$10^{-6.7}$	$10^{-10.2}$	5.7	$10^{-5.1}$
10^{-5} M NaHCO_3	$10^{-8.3}$	$10^{-5.7}$	10^{-5}	10^{-5}	$10^{-9.3}$	6.3	$10^{-5.1}$
$10^{-5}\text{ M Na}_2\text{CO}_3$	$10^{-8.3}$	$10^{-5.7}$	10^{-5}	$10^{-5.7}$	$10^{-8.3}$	6.6	$10^{-5.1}$

Water Chemistry Snoeyink And Jenkins Solutions Manual

Daniela Niemeyer



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Water Chemistry Vernon L. Snoeyink, David Jenkins, 1991-01-16 A first level text stressing chemistry of natural and polluted water and its application to waste water treatment Discusses principles of chemical kinetics dilute solution equilibria effects of temperature and ionic strength and thermodynamics in relation to water chemistry Strong emphasis given to graphical procedures Contains numerous example problems *Water Treatment Unit Processes* David W.

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Internal Corrosion of Water Distribution Systems, 2 Edition American Water Works Association,Awwa,1996-06

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valuable and comprehensive source of information on the rapidly evolving field of P recovery and recycling engineering for students researchers and professionals responsible for sustainable use of phosphorus The Publishers' Trade List Annual ,1981 Geochemical and Biogeochemical Reaction Modeling Craig M. Bethke,2022-01-06 Comprehensive primer handbook on geochemical reaction modeling from its origins and theoretical underpinnings to fully worked examples *British Books in Print* ,1979 Geochemical Reaction Modeling Craig Bethke,1996 An overview of the use of numerical methods to model reaction processes in the Earth s crust and on its surface The theoretical foundations of the field are discussed together with examples and case studies demonstrating the techniques that can be applied to scientific and practical problems Water & Sewage Works ,1973-02 **Journal** ,1981 Scientific and Technical Books and Serials in Print ,1984 *Control of Disinfection By-products in Drinking Water Systems* Anastasia Nikolau,Luigi Rizzo,Hüseyin Selcuk,2007 The occurrence of disinfection by products DBPs in drinking water has been an issue of major concern during several decades The formation of many DBPs species during water disinfection has been documented while new by products are still being detected as the analytical instrumentation available becomes more accurate and sensitive Most of the DBPs have been proven to have toxic effects on living organisms therefore they pose risks to human health during drinking water consumption The factors affecting their formation have been extensively investigated their transport and fate have been studied modelling efforts for several of them have been performed in order to understand better their behaviour and therefore try to minimise their occurrence in waters Techniques for their removal from water have also been applied and a variety of disinfection methods or combinations of disinfecting agents have been investigated with the aim to produce safe drinking water containing the minimum possible concentrations of DBPs This book deals with the advances in control of DBPs in drinking water systems Further than an providing an overview of existing disinfection techniques and by products up to date information on the parameters affecting the procedures of DBPs formation analytical methods for their determination toxicity regulation it pays special emphasis on the advanced treatment methods applied recently for DBPs control and presents recent promising findings as well as case studies in this field as the relevant research is proceeding producing more knowledge and practical solutions in regard to the disinfected drinking water quality **Metals Speciation Separation and Recovery** James W. Patterson,1987-07-05 This comprehensive book covers metals chemistry separation chemistry and metals separation processes State of the art papers give new and recent developments and future research needs **Water Chemistry** Vernon Snoeynik,1996-04-01 Choice ,1980 National Union Catalog ,1980 Includes entries for maps and atlases **Book of Abstracts** American Chemical Society. Meeting,American Chemical Society,1973

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