

CHEMISTRY 11 – WORD EQUATIONS AND TYPES OF CHEMICAL REACTIONS WORKSHEET

- 1) Translate the following word equations into chemical symbols and balance the resulting equations. Do not include the phases.

(a) Potassium + water $\rightarrow$ potassium hydroxide + hydrogen



(b) Nitrogen + hydrogen $\rightarrow$ ammonia



- 2) In each of the following write a balanced chemical equation, including the phases.

(a) Liquid phosphoric acid reacts with aqueous barium hydroxide to give water and a precipitate of barium phosphate



(b) Solid aluminium oxide and aqueous sulphuric acid produce water and aqueous aluminium sulphate



(c) Nitrogen trifluoride gas and hydrogen gas react to form nitrogen gas and gaseous hydrogen fluoride



(d) Powdered sodium carbonate and aqueous hydrobromic acid react to form carbon dioxide gas, aqueous sodium bromide and water



- 3) Classify the following reactions.

Reaction	Type
$2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$	Combustion
$\text{Mg} + \text{CuSO}_4 \rightarrow \text{MgSO}_4 + \text{Cu}$	Single Replacement
$4\text{Na} + \text{O}_2 \rightarrow 2\text{Na}_2\text{O}$	Synthesis
$\text{Sn}(\text{OH})_4 + 4\text{HBr} \rightarrow 4\text{H}_2\text{O} + \text{SnBr}_4$	Neutralization
$2\text{Fe}(\text{NO}_3)_3 + 3\text{MgS} \rightarrow \text{Fe}_2\text{S}_3 + 3\text{Mg}(\text{NO}_3)_2$	Double Replacement
$2\text{N}_2\text{O} \rightarrow 2\text{N}_2 + \text{O}_2$	Decomposition

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Roman Wölfel



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Chemical Literacy and Writing Chemical Reactions Nikolay Gerasimchuk, 2022-09-12 Writing chemical reactions in general and inorganic chemistry is not a trivial task. However, writing reactions for chemical processes correctly is a clear indicator of proficiency and competence in a subject. Unfortunately, very few students grasp the concept of the correct writing of chemical reactions quickly and so are unable to move through topics of general analytical and inorganic chemistry freely. Because the ability to write and balance different types of chemical reactions is a fundamental issue, this becomes a key question of chemical literacy. The successful writing of chemical reactions includes two components: the prediction of products of these reactions and their possible variations and balancing these reactions providing a material balance between starting compounds and reactions products. This book explores that element of the teaching of the fundamentals of chemical literacy: writing complete equations of chemical reactions and balancing them. It contains 49 figures, 22 schemes and 12 tables and 93 problems with answers. This book will be very useful for high school students interested in chemical sciences, higher education teachers, students in colleges and universities majoring in chemistry and biochemistry, and chemistry professionals working in industry. It also contains information about properties of the most common elements and applications of a variety of their chemical compounds.

A Pronouncing Chemical Formula Speller and Contest Guide Carl Alfred Jacobson, 1925

Cracking the AP Chemistry Exam Paul Foglino, 2010-08 Provides techniques for achieving high scores on the AP chemistry exam and includes two full length practice tests

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Basics for Chemistry David A. Ucko, 2013-09-24 Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of chemical change and nuclear and organic chemistry Undergraduate students will find the book as a very valuable academic material

Chemistry Equations And Answers (Speedy Study Guides) Speedy Publishing, 2015-04-24 Chemistry is a difficult subject to fully comprehend with its equations and scientific laws Trying to digest an entire book in one semester is a tough job but with the help of study guides like these you can absorb information in chemistry much more effectively This guide covers chemical equations including examples potential problems and solutions

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