

BALANCING CHEMICAL EQUATIONS

Name Grove

Rewrite and balance the equations below.

- $\text{N}_2 + \text{H}_2 \rightarrow \text{NH}_3$ $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
- $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$ $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$
- $\text{NaCl} + \text{F}_2 \rightarrow \text{NaF} + \text{Cl}_2$ $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$
- $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- $\text{AgNO}_3 + \text{MgCl}_2 \rightarrow \text{AgCl} + \text{Mg}(\text{NO}_3)_2$ $2\text{AgNO}_3 + \text{MgCl}_2 \rightarrow 2\text{AgCl} + \text{Mg}(\text{NO}_3)_2$
- $\text{AlBr}_3 + \text{K}_2\text{SO}_4 \rightarrow \text{KBr} + \text{Al}_2(\text{SO}_4)_3$ $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow 6\text{KBr} + \text{Al}_2(\text{SO}_4)_3$
- $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
- $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{C}_3\text{H}_8 + 5\text{O}_2 \rightarrow 3\text{CO}_2 + 4\text{H}_2\text{O}$
- $\text{C}_8\text{H}_{18} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{C}_8\text{H}_{18} + \frac{25}{2}\text{O}_2 \rightarrow 8\text{CO}_2 + 9\text{H}_2\text{O}$
- $\text{FeCl}_3 + \text{NaOH} \rightarrow \text{Fe}(\text{OH})_3 + \text{NaCl}$ $\text{FeCl}_3 + 3\text{NaOH} \rightarrow \text{Fe}(\text{OH})_3 + 3\text{NaCl}$
- $\text{P} + \text{O}_2 \rightarrow \text{P}_2\text{O}_5$ $4\text{P} + 5\text{O}_2 \rightarrow 2\text{P}_2\text{O}_5$
- $\text{Na} + \text{H}_2\text{O} \rightarrow \text{NaOH} + \text{H}_2$ $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$
- $\text{Ag}_2\text{O} \rightarrow \text{Ag} + \text{O}_2$ $2\text{Ag}_2\text{O} \rightarrow 4\text{Ag} + \text{O}_2$
- $\text{S}_8 + \text{O}_2 \rightarrow \text{SO}_3$ $\text{S}_8 + 12\text{O}_2 \rightarrow 8\text{SO}_3$
- $\text{CO}_2 + \text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2$ $6\text{CO}_2 + 6\text{H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6\text{O}_2$
- $\text{K} + \text{MgBr}_2 \rightarrow \text{KBr} + \text{Mg}$ $2\text{K} + \text{MgBr}_2 \rightarrow 2\text{KBr} + \text{Mg}$
- $\text{HCl} + \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$ $2\text{HCl} + \text{CaCO}_3 \rightarrow \text{CaCl}_2 + \text{H}_2\text{O} + \text{CO}_2$

Writing And Balancing Chemical Equations Answers

**Morris Hein, Scott Pattison, Susan
Arena, Leo R. Best**



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National 5 Chemistry with Answers, Second Edition Barry McBride, Stephen Jeffrey, John Anderson, Paul McCranor, Fran Macdonald, 2018-07-30 Exam Board SQA Level National 5 Subject Chemistry First Teaching September 2017 First Exam Summer 2018 The second edition of this textbook has been fully revised and updated to reflect changes made to the SQA syllabus from 2017 onwards New features include Refreshed content Additional candidate advice Model answers for open ended questions

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