

Total No. of Questions : 6]

SEAT No. :

P462

[Total No. of Pages : 3

[5115] - 205
S.Y. B.Sc. (Semester - II)
CHEMISTRY
CH - 221: Physical and Analytical Chemistry
(2013 Pattern)

Time : 2 Hours]

[Max. Marks : 40

Instructions to the candidates:

- 1) All questions are compulsory.
- 2) Neat diagram must be drawn wherever necessary.
- 3) Figures to the right side indicate full marks.
- 4) Use of calculator is allowed.
- 5) Answer to both sections should be written in separate answer sheet.

SECTION - I
[Physical Chemistry]

Q1) Answer the following. [5]

- a) State the criteria of equilibrium in terms of ΔF .
- b) Write the Gibbs - Helmholtz equation.
- c) Draw the T-N diagram of solution with negative deviation.
- d) Define the term: Lower critical solution temperature.
- e) Define the term: Normality.

Q2) a) Attempt any two of the following. [6]

- i) Derive the Clapeyron equation.
- ii) State and explain Henry's law. Give its applications.
- iii) Explain with diagram the mutual solubility behaviour of phenol - water system at various temperatures.

b) Answer any two of the following. [4]

- i) State Raoult's law and give its equation.
- ii) Explain the physical significance of ΔF .
- iii) How will you calculate graphically the heat of vaporisation of liquid from Clausius - Clapeyron equation.

P.T.O.

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schäferei worauf der heutige straßenname schäferenberg zurückgeht als der lugheim wird im 19 jahrhundert die anhöhe zwischen seeligstadt und taubenheim bezeichnet weblinks bearbeiten quelltext bearbeiten

die lügen geschichte wer die welt wirklich regiert - Sep 04 2022

web wer regiert die welt sonderausgabe ian morris 2012 08 gibt es einen roten faden durch die geschichte der uns im rückblick zeigt wohin die zukunft uns führt ian morris hat ihn aufgerollt aus einer vielzahl historischer fakten archäologischer funde