ection 1 Atom 1. How are atoms and	s, Ions, and Molecu I elements related?	les	Hr: _	
Sketch the structur				
Label the protons,	neutrons, nucleus, and electro	ens.		
How do compound	k differ from elements?			
. How do compound	o direi non escircito.			
4. What is an ion?				
5. Why does an ion h	ave an electrical charge?			
5. In the spaces provi	ded below, sketch how both p	sositive and negative	e ions form. Label the nucleus	and the
	ire 1.3 as a reference.			
What is a covalent	bond?			
8. What determines the	ne number of covalent bonds to	that an atom can for	m?	
se the following wo	rds identify to the correct d	efinition below.	*	
element	compound	ion	molecule	
	9. atoms held together by	covalent bonds		
	10. composed of different ty			
	11. composed of one type o			
	12. atom that has gained or			
	12. anoth that has gamed or	BOSE CHOCHRONES		
3. What is the different	nce between how ionic and co	walent bonds from		
, is the unitere	the restricted from four and co	CONTRACTOR CONTRACTOR		
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