

Tips for naming molecules

There are three steps, though that I have found useful over the years. The first concerns organic molecules, the second, inorganic, the third, for polymers, simple ones.

Remember that the name of an organic molecule has three parts:

- **Numbering** (if necessary)
- **Base**: the parent chain (length and C/C bonding)
 - o First realise that some will be complicated.
- **Suffix**: substituents dictated by the highest priority functional group

1. Start off by identifying the highest priority functional group: this helps you determine the suffix of the molecule's name which can help you determine where you start numbering the parent chain. In subsequent paragraphs, however, I mention you begin to identify the parts of the molecule that will be added to the prefix.

If you only have C and H atoms left you have hydrocarbons that are based on parent hydrocarbons: the prefix is the key.

2. There begins to be the longest continuous carbon chain you can (the only thing is: branching is good)

- Make sure that you count the longest continuous carbon chain (highest priority functional group) that begins before you get to the first carbon.
 - o Carbonyls, nitriles, aldehydes, esters, amines, amides
- If you include the carbon that most carbon chains are the longest the result is a complete structure.
- Always include the functional groups and multiple bonds in the structure.

3. Begin to number the carbons. The lowest number that you can get that is not the highest is the lowest number for the following substituents (if you can't get it):

- The highest priority functional group
- Any multiple bonds
- The substituent closest to the end
- The complete structure (if you can't get it)

4. Once you have numbered the parent chain, you can now the appropriate name to indicate the longest chain chain.

5. Identify the other substituents (nitriles, aldehydes, esters, amines, amides, alcohols, alkenes, alkynes, etc.) and identify the other parts of the molecule that are not the parent chain (if you can't get it).

- Only C/C single bonds: "and"

Study Guide Naming Molecules

Steven S. Zumdahl, Paul B. Kelter



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