

10.1 Periodic Trends



10.1 PERIODIC TRENDS: FROM CONCEPT TO APPLICATION

Section 10.1 – continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	20
_____	Hydrogen
_____	Li (3)
_____	Ne (10)

Atomic Mass	22.99
Atomic Number	19
Electron Configuration	$1s^2 2s^2 2p^6 3s^1$
Chemical Name	Sodium
Chemical Symbol	Na

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 16. It exists as the periodic table.
- _____ 17. It has no the periodic table.
- _____ 18. Group 1 elements.
- _____ 19. Elements that are strong oxidizing agents.
- _____ 20. Group 2 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. transition elements

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word or phrase to make it true.

- _____ 21. There are two representative elements of elements.
- _____ 22. Most elements have higher of the elements in the periodic table are metals.
- _____ 23. Group 1A elements except for hydrogen are metals in the solid state.
- _____ 24. Group 2A elements are the alkaline earth metals.
- _____ 25. Group 1A elements are highly reactive elements known as alkali metals.
- _____ 26. Group 2A elements are very reactive elements known as alkaline earth metals.
- _____ 27. Alkaloids have properties of both metals and non-metals.

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