# CHEMISTRY (SCIENCE): STOICHIOMETRIC CALCULATIONS (MOLE) HARVARD UNIVERSITY Study guide: Definitions, Explanation, Examples and Notes

#### Calculations from

equations Mole to

mole calculations

#### Examples

 How many moles of chlorine are required to react 2.5 moles of calcium to produce calcium chloride?

#### Solution

$$Ca_{(s)} + Cl_{2(g)} \rightarrow CaCl_{2(s)}$$

$$x = \frac{2.5 \operatorname{mol} \operatorname{Ca} \times 1 \operatorname{mol} \operatorname{Cl}_2}{1 \operatorname{mol} \operatorname{Ca}}$$

$$x = 2.5 \text{ mol}$$

 How many moles of carbon dioxide (CO<sub>2</sub>) will be produced by complete combustion of 2 moles of glucose (C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>) according to the equation?

$$C_6H_{12}O_{6(sq)} + 6O_{2(g)} \rightarrow 6CO_{2(g)} + 6H_2O_{(l)}$$

#### Solution

$$1 \text{ mol } C_6H_{12}O_6 \rightarrow 6 \text{ mol } CO_2$$

$$2 \text{ mol } C_6H_{12}O_6 \longrightarrow x$$

$$x = \frac{2 \mod C_6 H_{12} O_6 \times 6 \mod C O_2}{1 \mod C_6 H_{12} O_6}$$

$$x = 12 \mod CO_2$$

#### Mole to mass

#### *calculations*

#### Examples

## **Study Guide For Content Mastery Stoichiometric Calculations**

Onno de Jong

#### **Study Guide For Content Mastery Stoichiometric Calculations:**

Chemical Calculations: an Educational Study of Stoichiometric Calculations in Secondary School Chemistry Onno de Jong,1990 You Can Do Chemistry Sunil Tanna,2018-11-19 A comprehensive guide to performing mole and stoichiometric calculations with numerous examples as well as questions and answers Covers calculations relating to solids solutions gases and electrolysis plus as limiting and excess reactants chemical yields atom economy and much more Fully up to date with the last international standards including the revised definition of mole which was agreed on November 16th 2018

Stoichiometric Calculations Lifelige, 2019 This lesson plan covers how to calculate moles and volume ratios for balanced chemical equations Introduction to Process Calculations Stoichiometry KA. Gavhane, 2012 Factors in the Performance of Stoichiometric Calculations in Chemistry Ronald Zachariah Fernandez,1990 A Study Investigating a Hierarchical Model Relating to Stoichiometric Calculations Patrick J. Whelan, 1982 Students' Conceptions of Stoichiometry at the Submicro Level Sin-Yan Chan, [[]], 2017-01-27 This dissertation Students Conceptions of Stoichiometry at the Submicro Level by Sin van Chan was obtained from The University of Hong Kong Pokfulam Hong Kong and is being sold pursuant to Creative Commons Attribution 3 0 Hong Kong License The content of this dissertation has not been altered in any way We have altered the formatting in order to facilitate the ease of printing and reading of the dissertation All rights not granted by the above license are retained by the author Abstract Stoichiometry is an important topic in chemistry It tells how many reactants are required to produce a certain amount of product in terms of mass mole and volume Learning stoichiometric calculation involves the understanding of certain concepts such as the mole stoichiometric ratios and chemical equations Some studies attributed the failure in learning stoichiometry to the unfamiliarity with the amount of substance in a mole and students weakness in the mathematical ability Nevertheless how students connect the submicro level and the symbolic level in learning stoichiometry was not discussed widely In this study two examples of chemical reactions with different levels of difficulties were used to probe students conceptual understanding in stoichiometry at the submicro level Their strategies used in stoichiometric calculations were examined by an interview study of five Secondary Five students The connection between the submicro level and the symbolic level in learning stoichiometry would be also probed into Results indicate that the failure of stoichiometry learning may due to the disconnection in different levels of representation and students generated strategy one portion reasoning An implication for teaching and learning is that teachers should use diagrams at the submicro level in the teaching of stoichiometry Such diagrams should aim to help students building connections across the three levels of representation and enhancing students conceptual understanding in stoichiometry DOI 10 5353 th b5396387 Subjects Chemistry Study and teaching Secondary Stoichiometric Calculations: Known Problems and Proposed Solution at a Chemistry-mathematics Interface Werner Dierks, 1981 A Study Investigating a Hierarchical Model Relating to Stoichiometric Calculations Patrick J. Whelan, 1982

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